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Near-ambient pressure XPS for in-situ studies of catalytic surfaces

Dr. James N. O'Shea

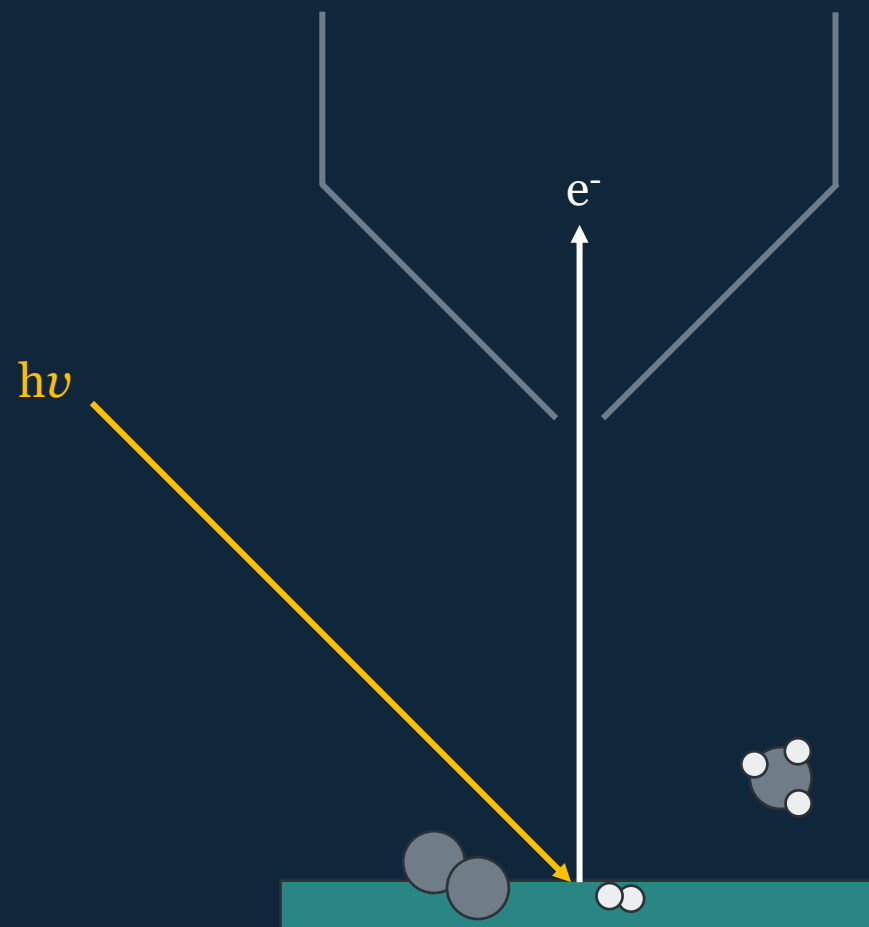
School of Physics & Astronomy, University of Nottingham, UK



**2nd LATAM MEETING ON GREEN
AMMONIA AND POWER-to-X**



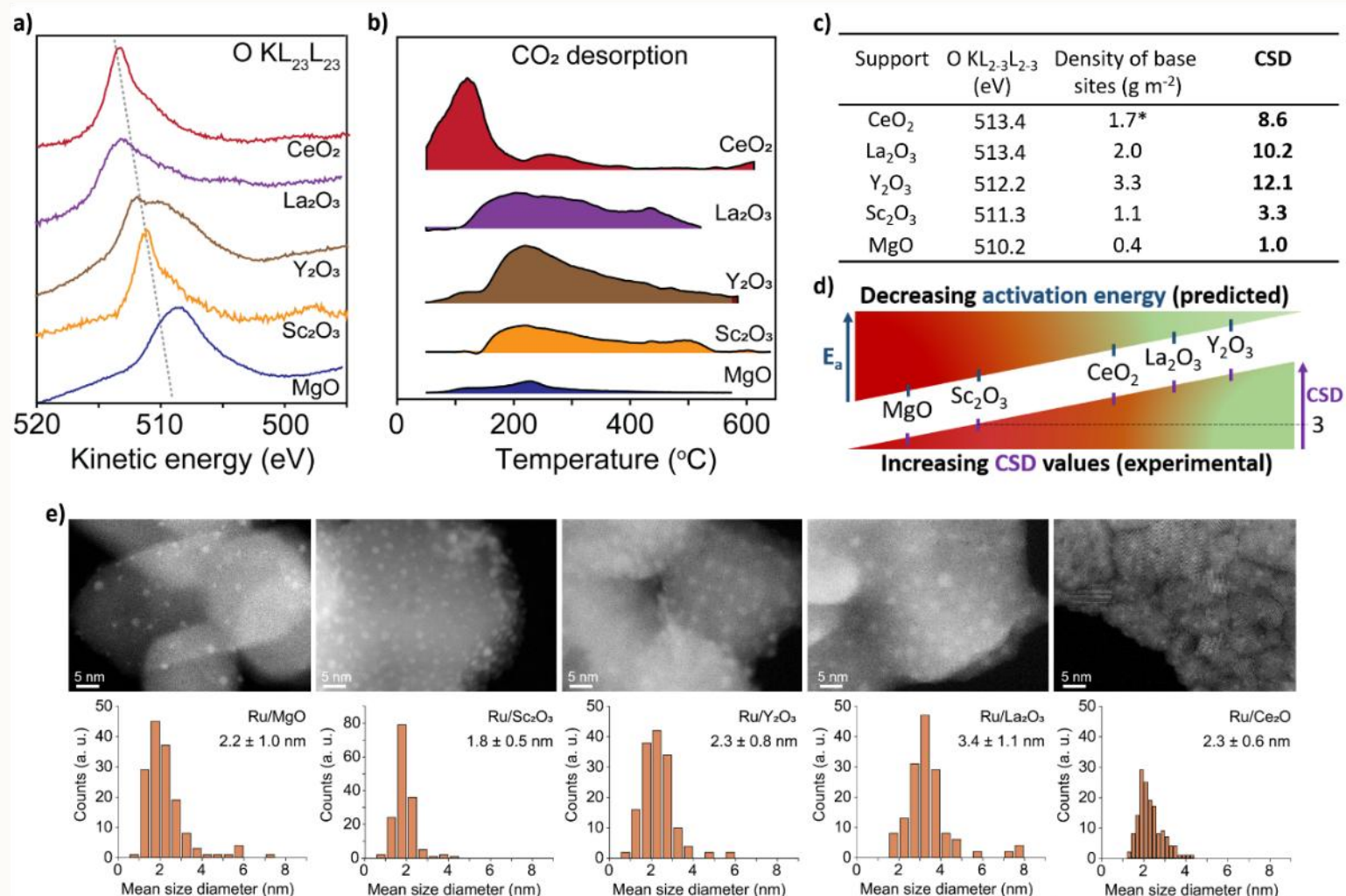
XPS & it's applications to catalytic surfaces





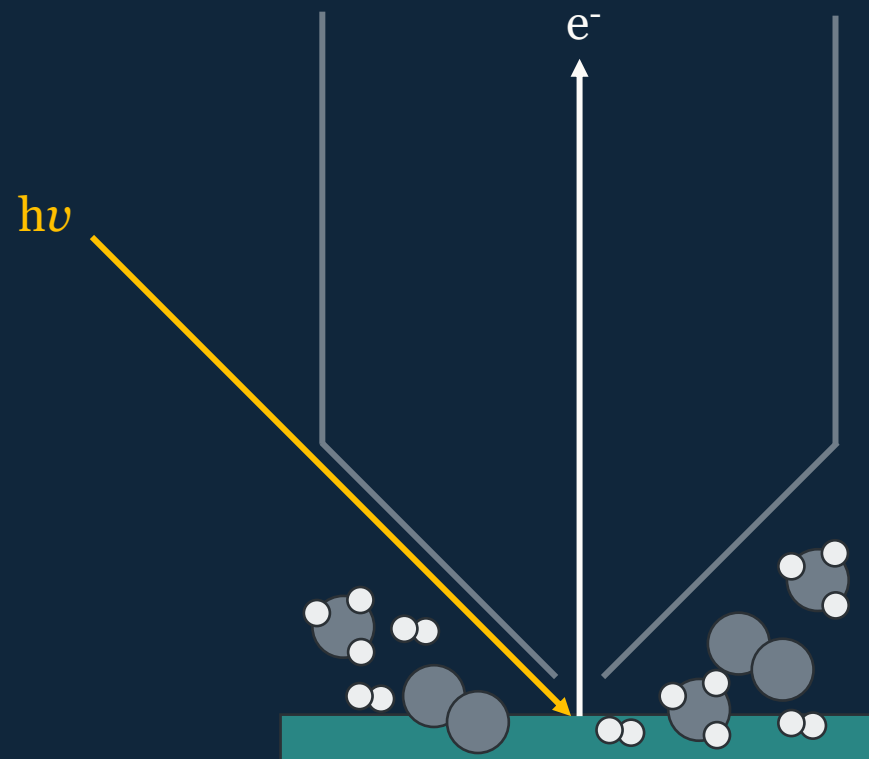
A Sole Descriptor Guiding the Selection of Catalyst Supports for Ammonia Synthesis

A. Weilhard, I. Popov, E. Kohlrausch, G. Aliev, L. S. Blankenship, L. T. Norman, S. Ghaderzadeh, L. Smith, M. Isaacs, J. N. O'Shea, A. Lanterna, W. Theis, D. Morgan, G. Hutchings, E. Besley, A. N. Khlobystov and J. Alves Fernandes



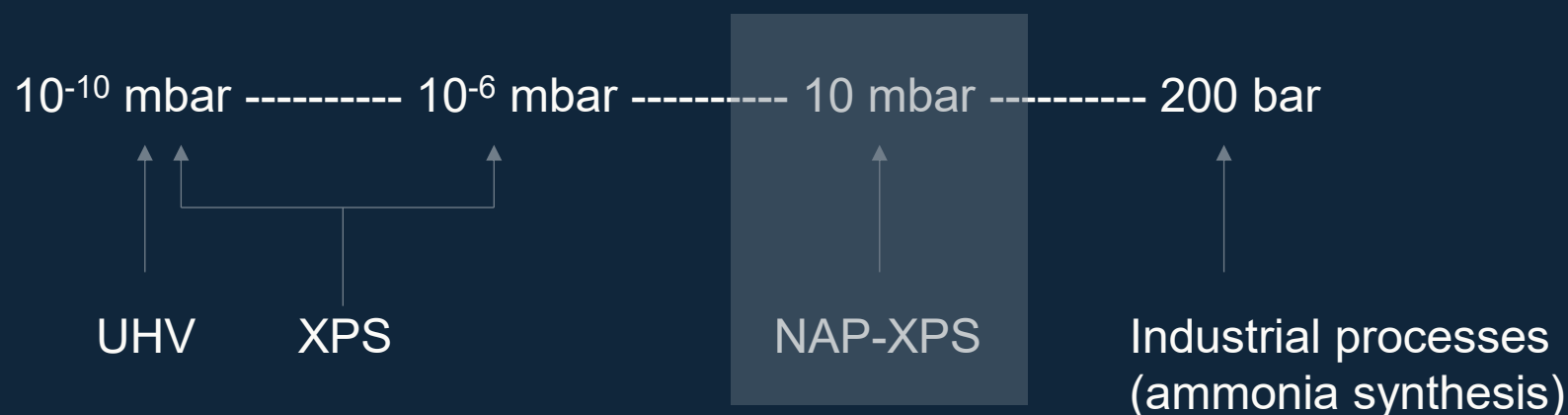


XPS & reaching for more realistic pressures





What does near-ambient pressure mean in the context of XPS?





Operando probing of the surface chemistry during the Haber–Bosch process

<https://doi.org/10.1038/s41586-023-06844-5>

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Christopher M. Goodwin^{1,5}✉, Patrick Lömker¹, David Degerman¹, Bernadette Davies², Mikhail Shipilin¹, Fernando Garcia-Martinez³, Sergey Koroidov¹, Jette Katja Mathiesen¹, Raffael Rameshan⁴, Gabriel L. S. Rodrigues¹, Christoph Schlueter³, Peter Amann^{1,6} & Anders Nilsson¹✉

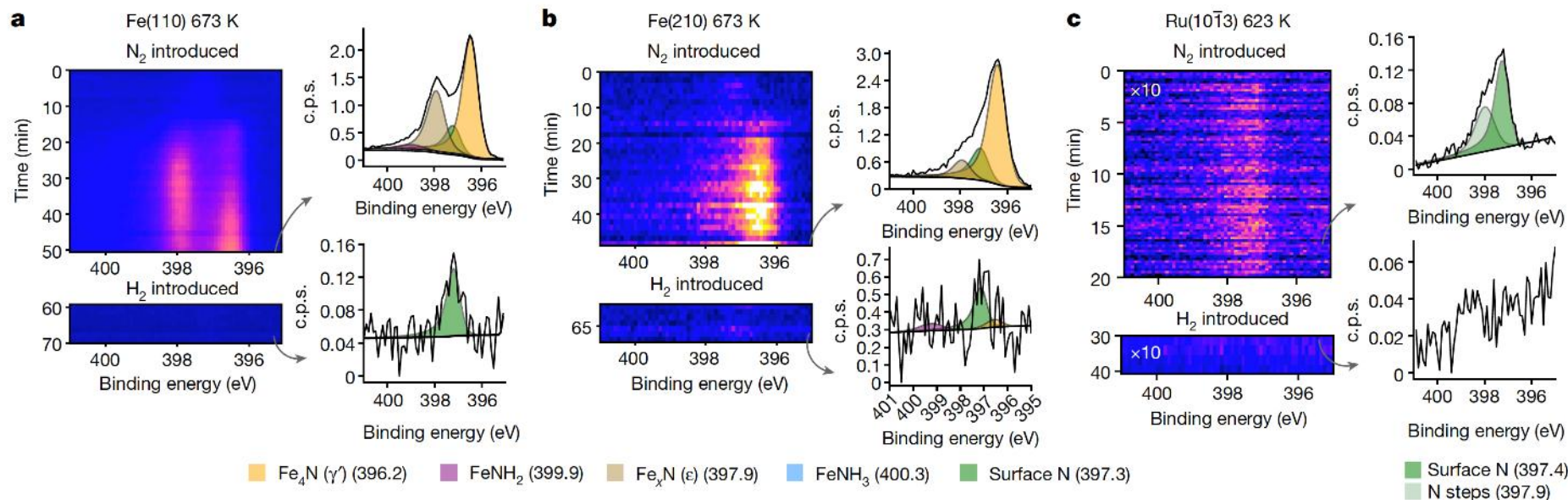


Fig. 2 | Nitride formation and depletion. The formation and depletion of nitride on the surface of each catalyst are shown as a function of time. At the top, the N₂ gas is introduced with a total pressure of 150 mbar and spectral collection begins. Then, after the nitride begins to stabilize, H₂ gas is introduced immediately in a 1:1 ratio with N₂ with a total pressure of 300 mbar, reducing the surface

within the frame of the detector. Next to each time series are example spectra normalized to the background, with a grey arrow showing the frame it represents. **a**, The data for 673 K over Fe(110). **b**, The data for 673 K over Fe(210). **c**, The data for 623 K over Ru(10 $\bar{1}$ 3). For Ru, the spectra shown are the summation of the entire time series. Note the difference in y-axis scale in the spectral figures.

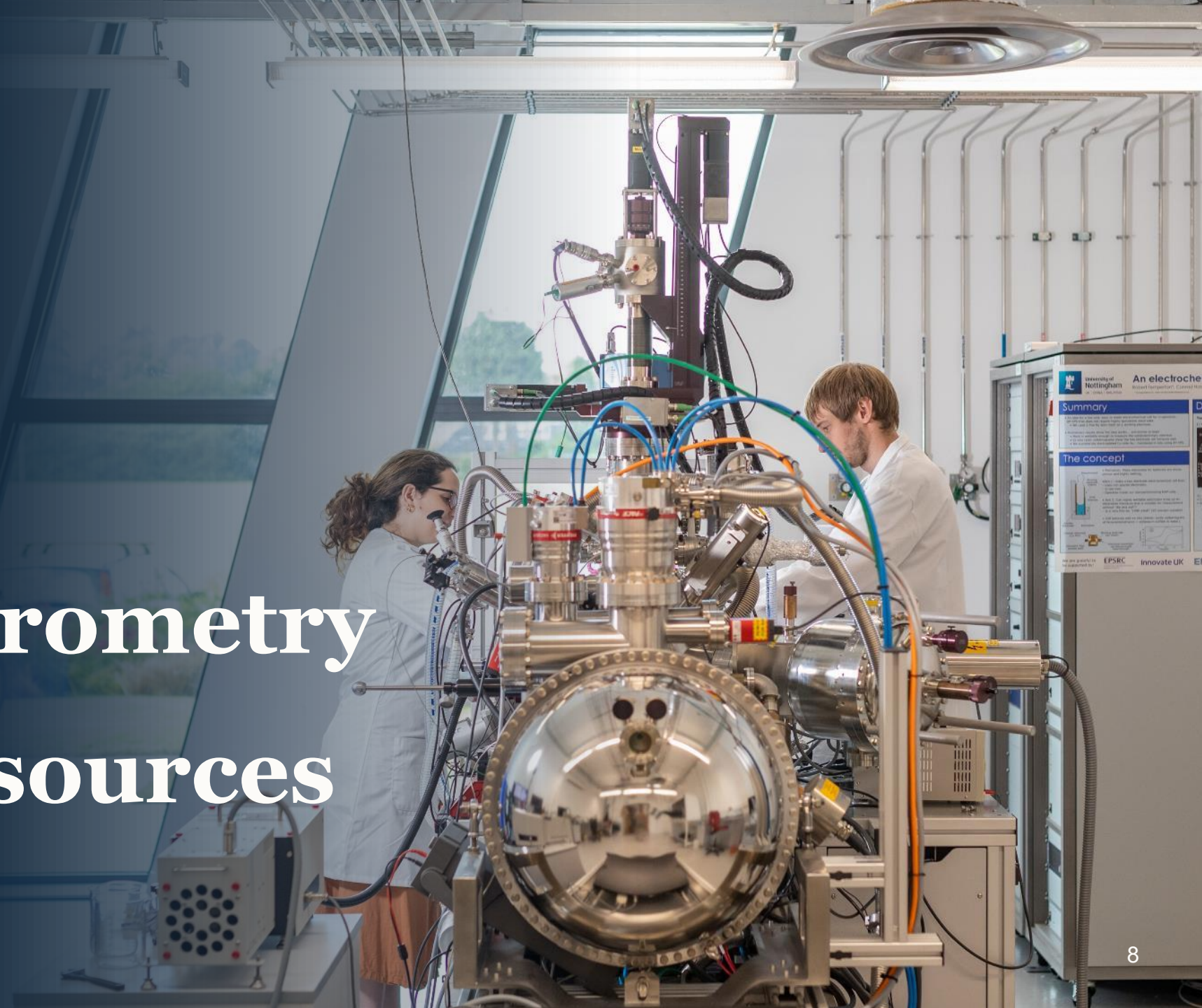


The instrument

NAP-XPS



XPS Gas-dosing Mass-spectrometry Excitation sources



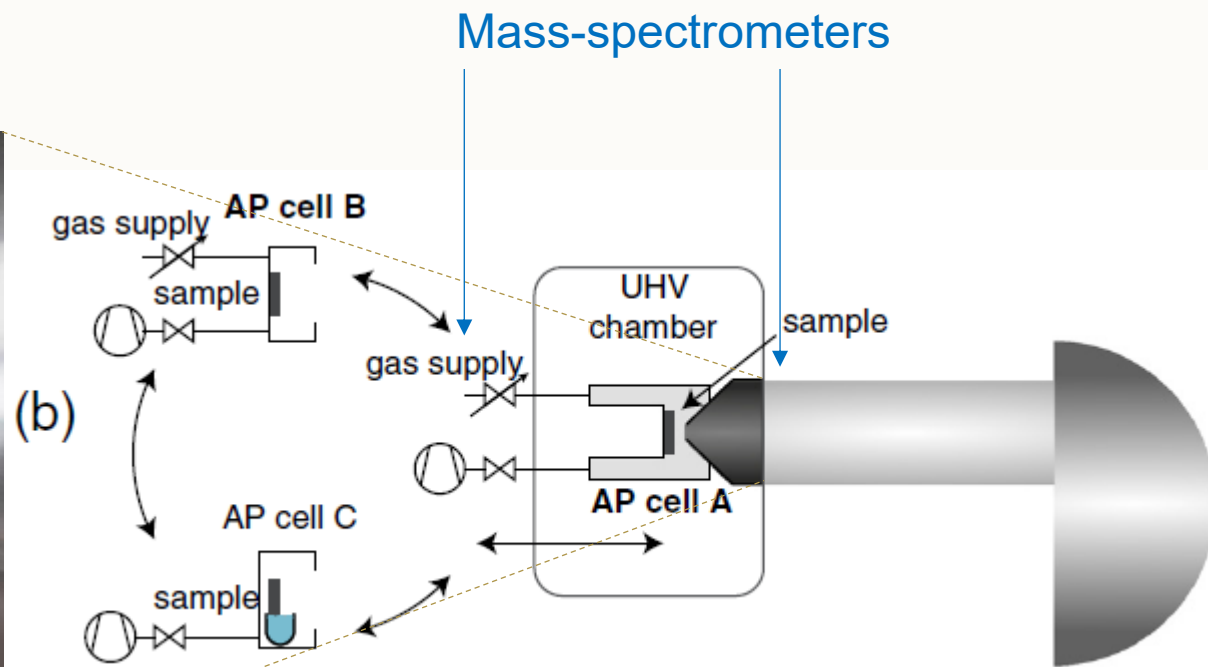


Bridging the pressure gap



X-rays

Visible & UV



UHV suitcase for sample transfer to other instruments



“Knowing the before-reaction part or after-reaction part is like studying a life with access only to the prenatal and postmortem states.”

- Gabor A. Somorjai, 1998

PAPER

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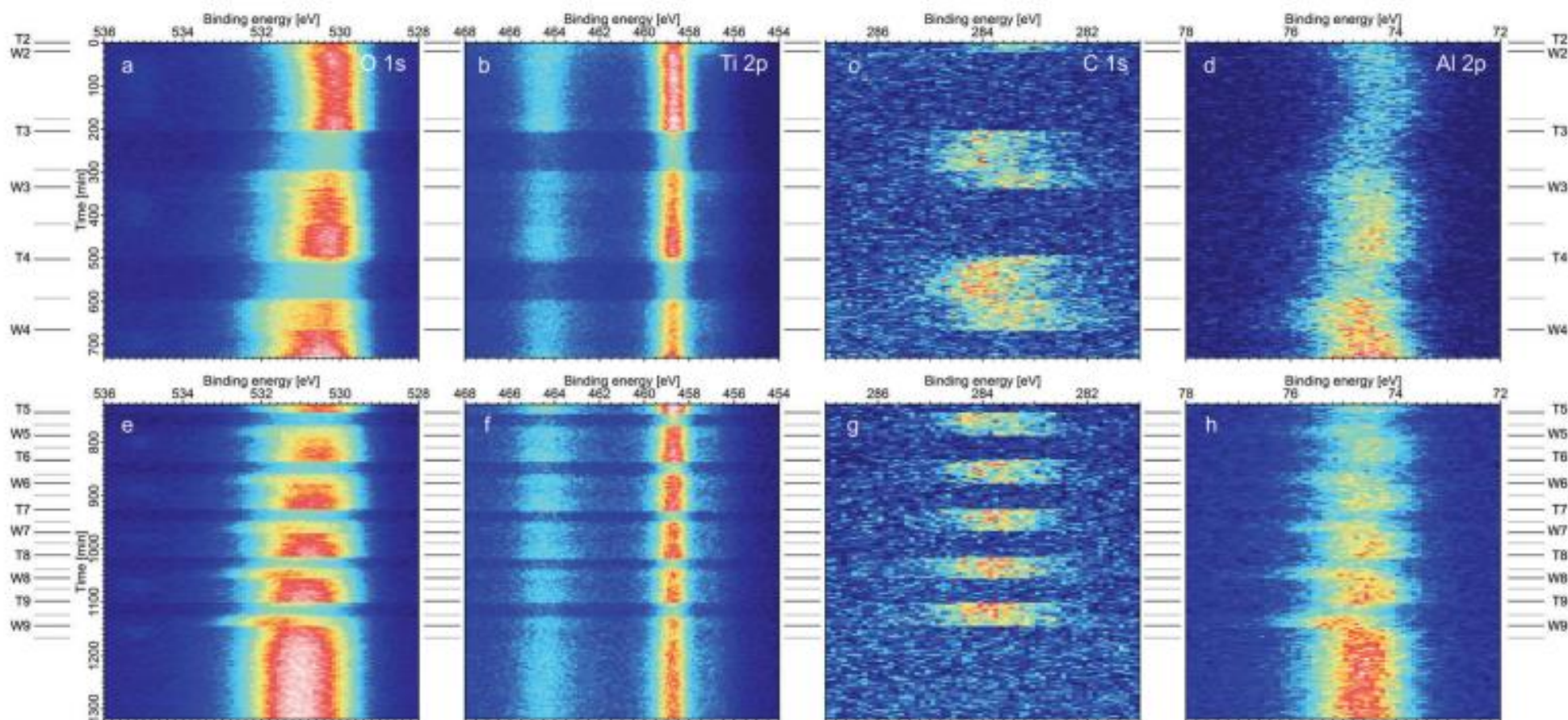
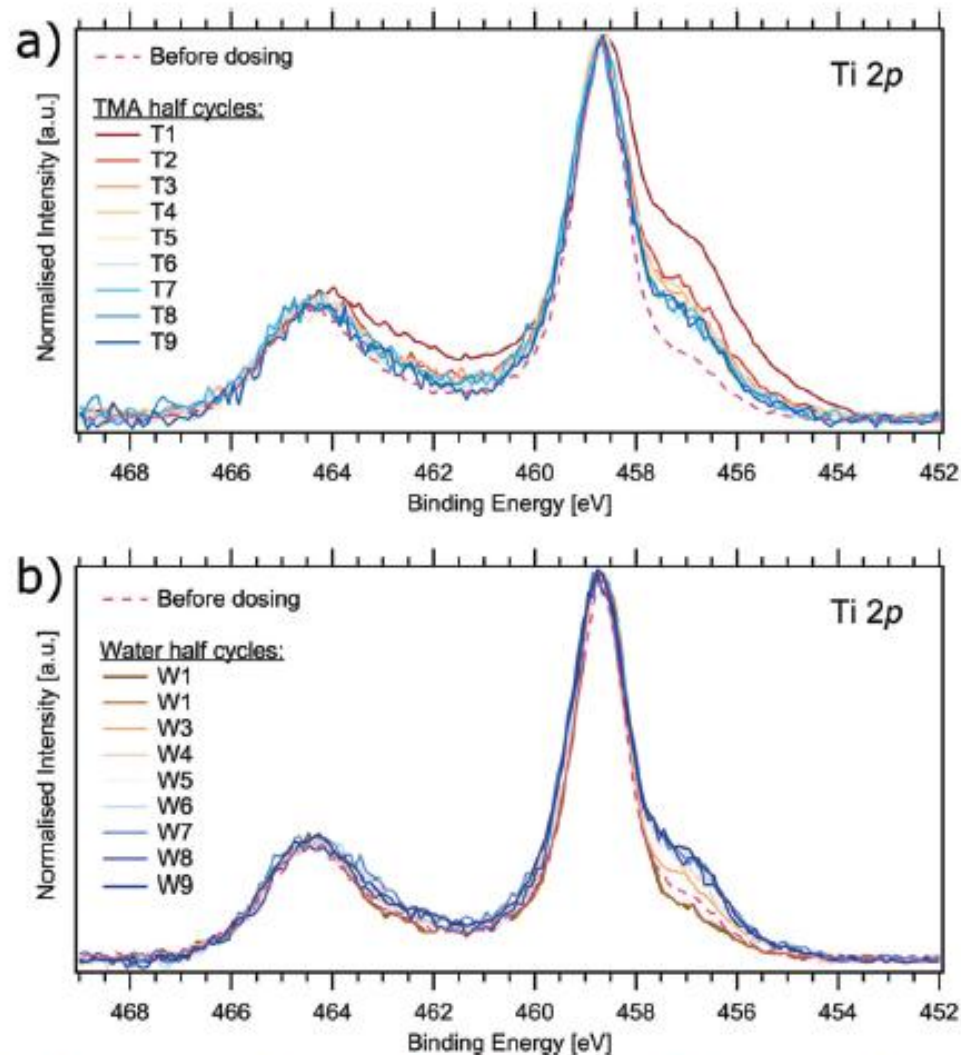


Fig. 2 *In situ* XPS measurements during cycles 2–9, labeled T2–T9 and W2–W9 for TMA and water half-cycles respectively. Grey lines indicate purges of the cell to 1×10^{-2} mbar. From left to right, O 1s, Ti 2p, C 1s and Al 2p regions are shown. The break in the data (between plots a–d and e–h) represents a break in the experiment, apart from which the measurements were continuous. Each cycle of the four regions took approximately 3 minutes 45 seconds to measure.



Realtime ALD of Al_2O_3 growth on TiO_2



Ti^{3+} defects are created at the interface due to the reaction of the surface with trimethyl aluminium

Ti^{3+} defects are healed by water exposure, but get locked-in at the interface after a full monolayer is formed.

Fig. 3 Ti 2p spectra after each TMA (a) and water (b) half cycles that have been normalised to the height of the $\text{Ti}_{3/2}^{4+}$ peak. A spectrum prior to dosing is included for reference.



An in situ exploration of subsurface defect migration to a liquid water-exposed rutile $\text{TiO}_2(110)$ surface by XPS

M. H. Mesbah Ahmed

School of Physics & Astronomy, University of Nottingham, Nottingham, UK

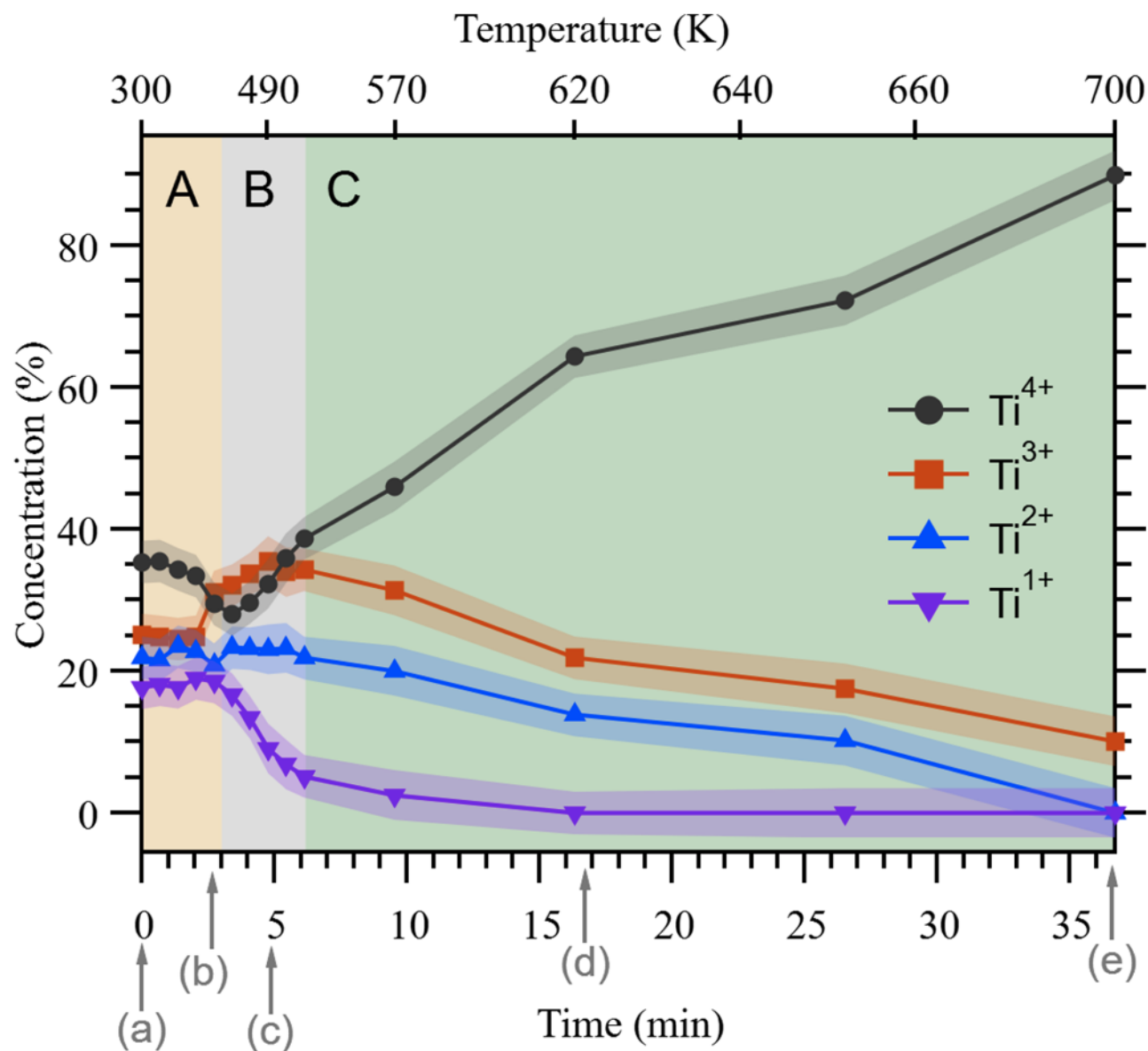
Correspondence

James N. O'Shea, School of Physics & Astronomy, University of Nottingham, Nottingham NG7 2RD, UK.
Email: J.Oshea@nottingham.ac.uk

Funding information

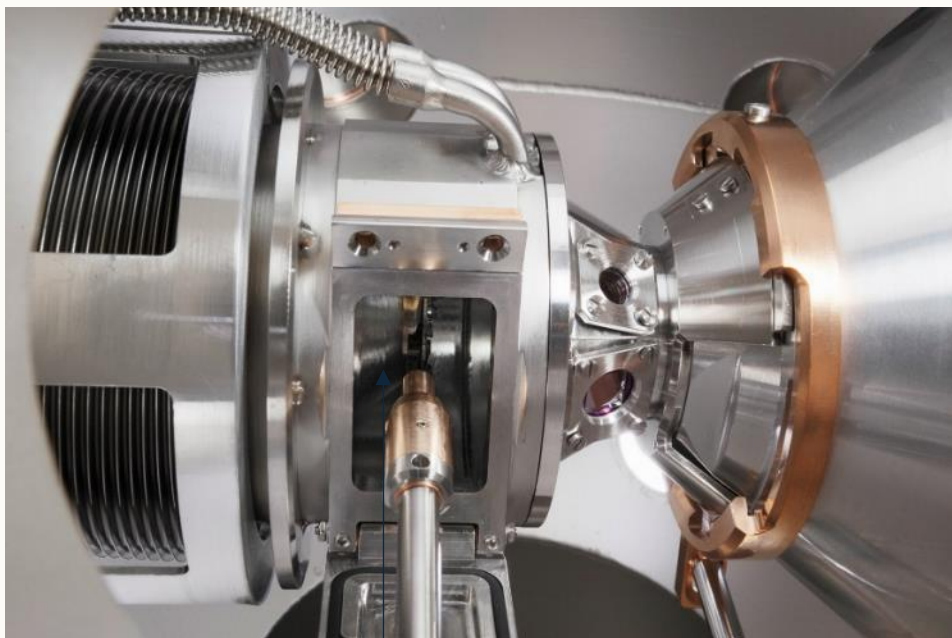
Engineering and Physical Sciences Research Council, Grant/Award Number: Doctoral Prize; Innovate UK, Grant/Award Number: ERA; University of Nottingham, Grant/Award Number: Propulsion Futures Beacon

Paper presented at the UK Surface Analysis Forum 40th Anniversary Meeting, University of Nottingham UK, 10-11 July 2019.

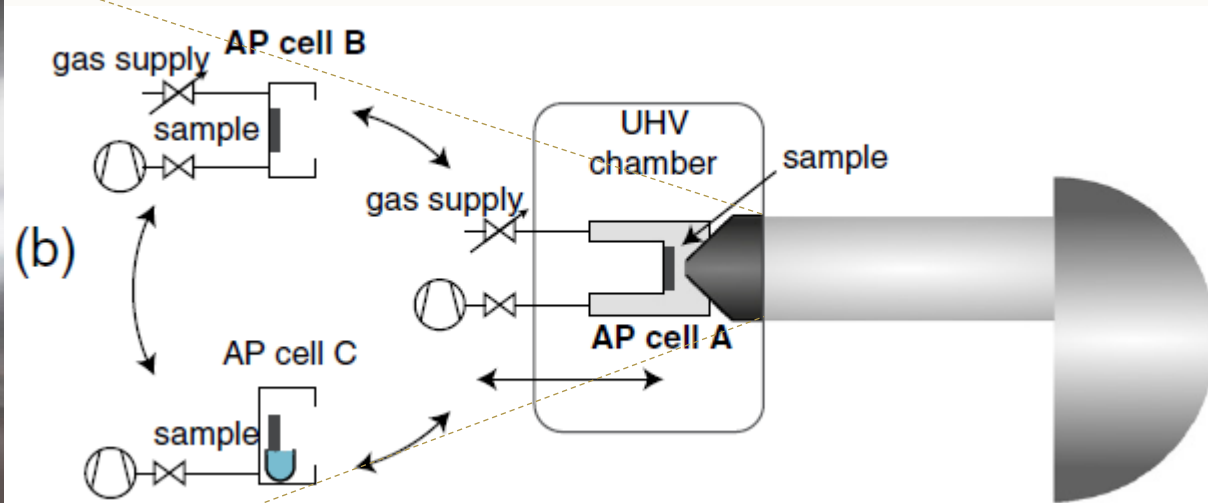




How do we get liquid water on the surface?

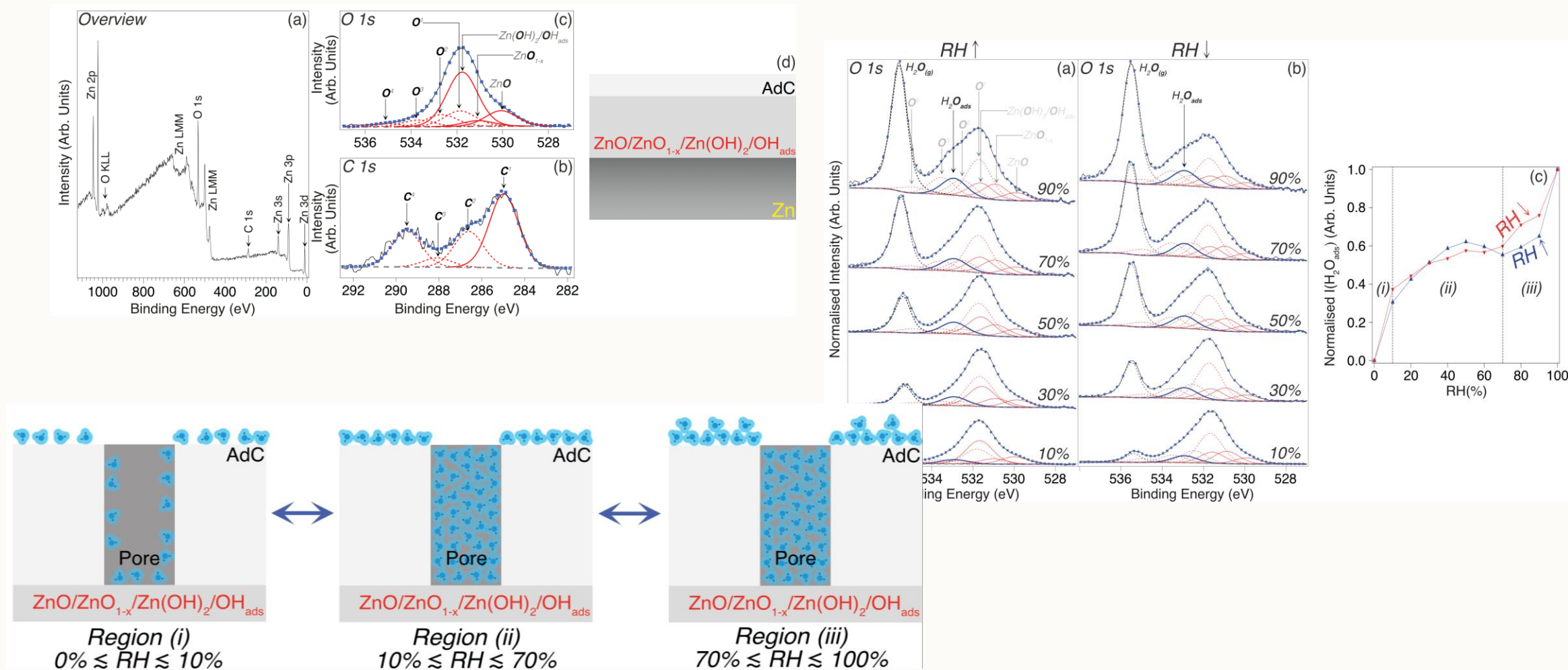


Peltier thermoelectric cooled sample





Complexity at a Humid Surface: Throwing Light on Atmospheric Corrosion





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The growth of GaSe & oxidation to Ga_2O_3

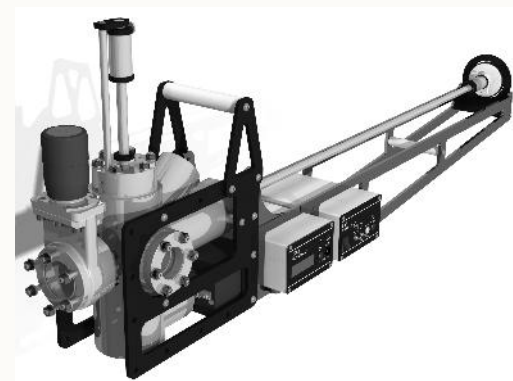
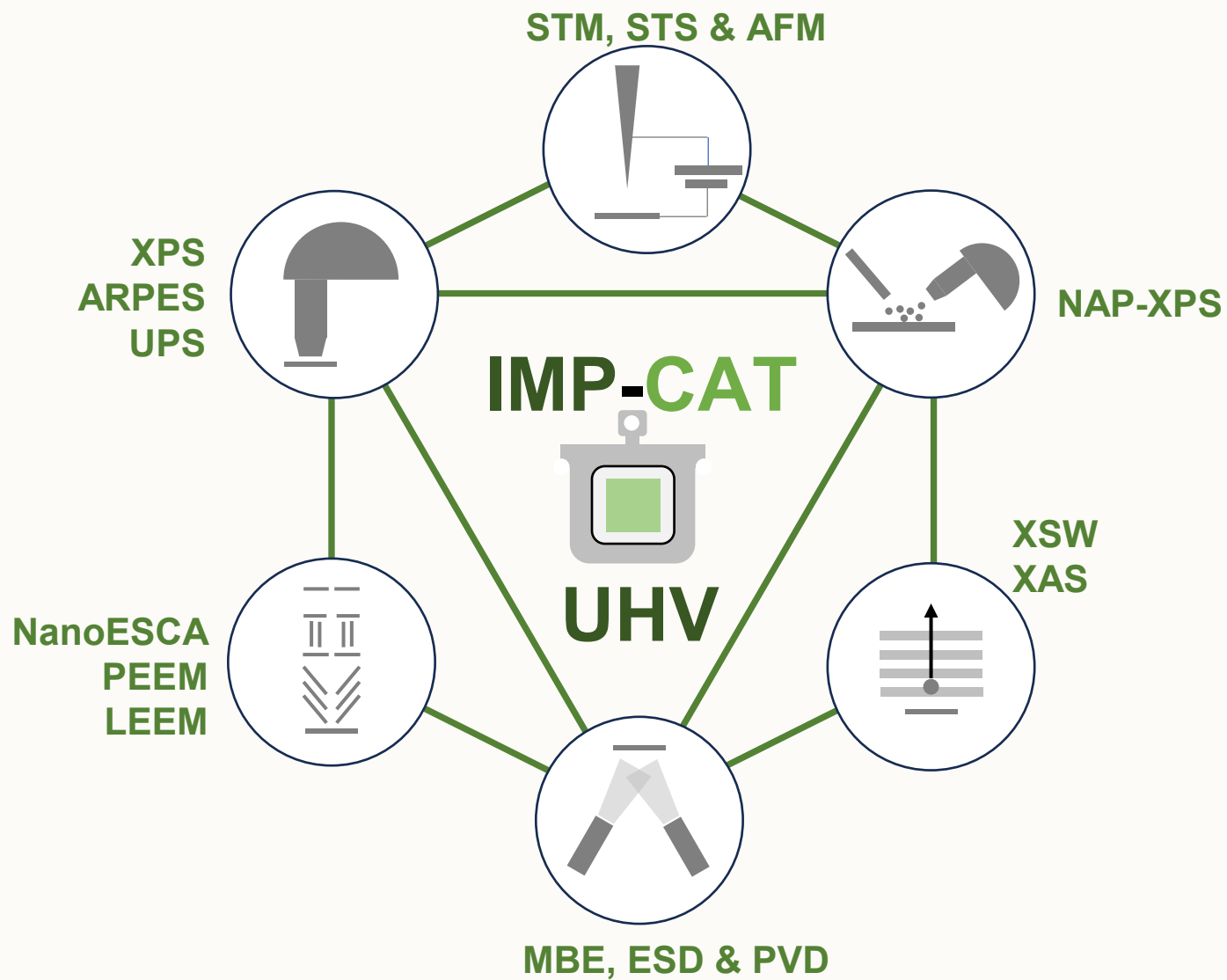
EPI2SEM & NAP-XPS

IMP-CAT



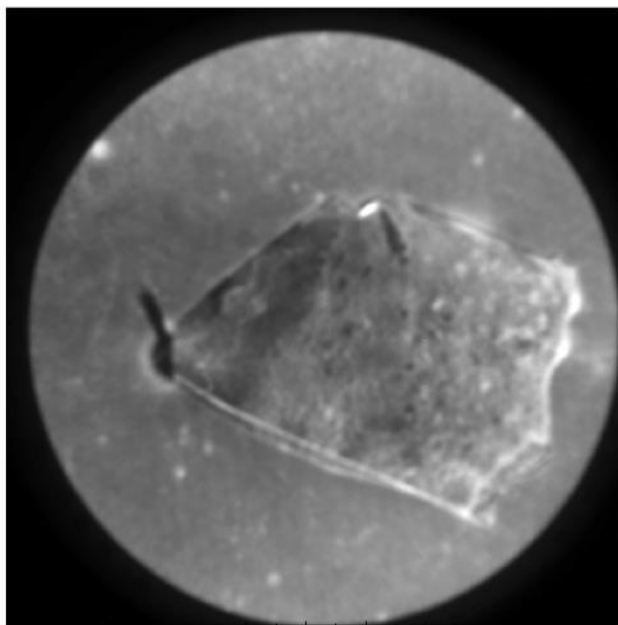
IMP-CAT

Interconnected Materials Preparation & Characterisation for Advanced Technologies

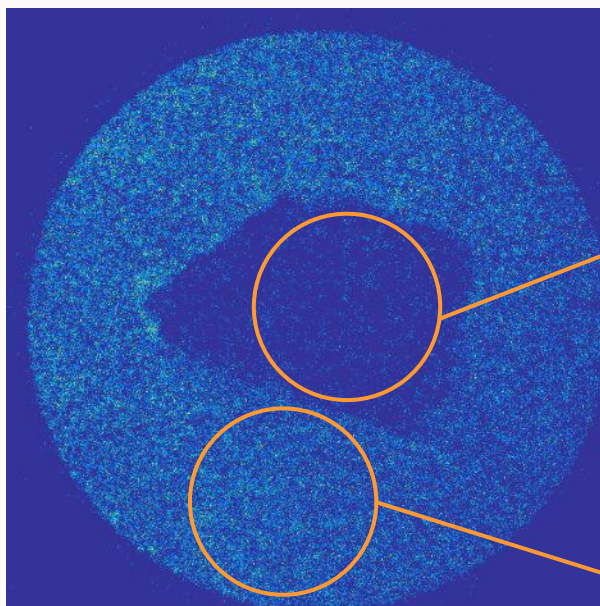




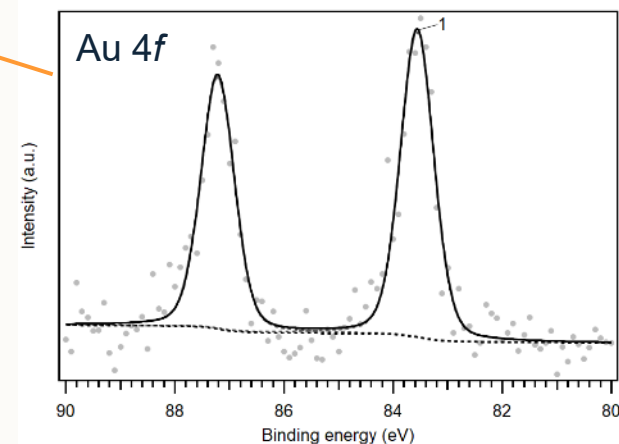
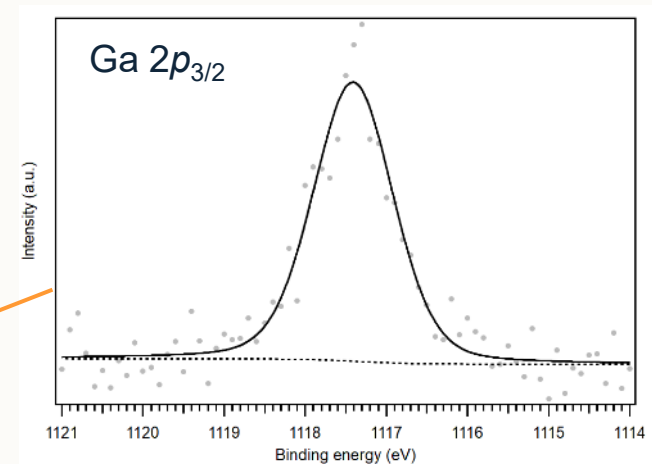
XPEEM of a 20 μm GaSe flake on Gold



Hg-lamp image

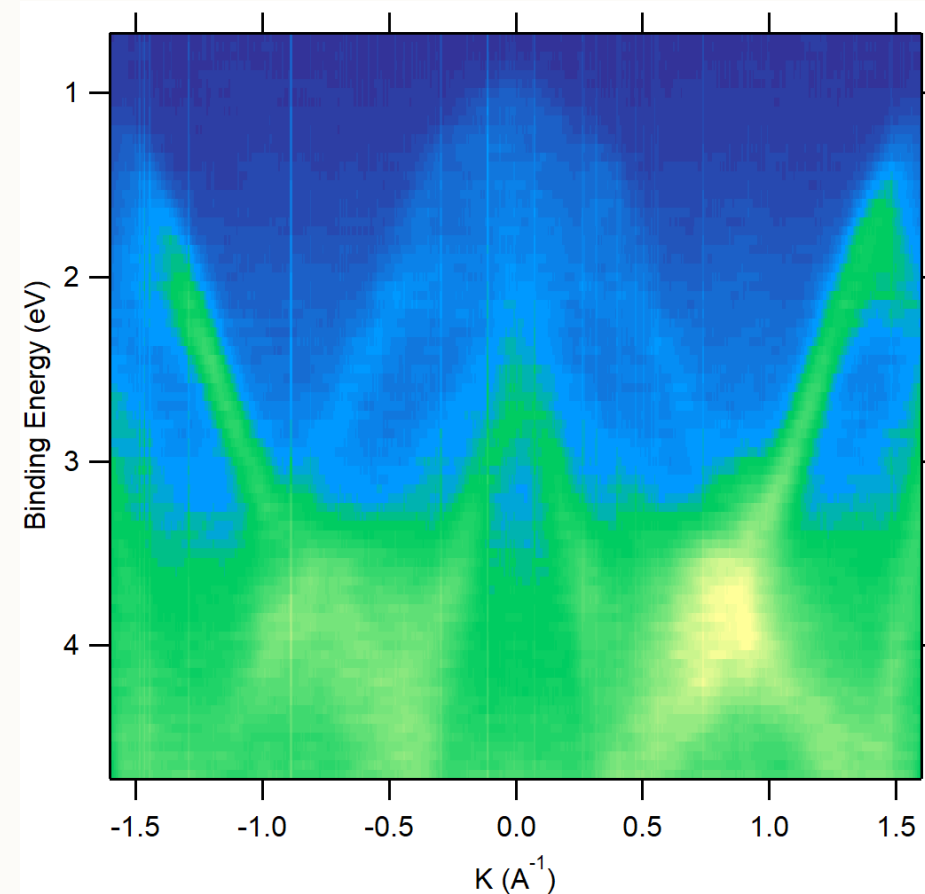
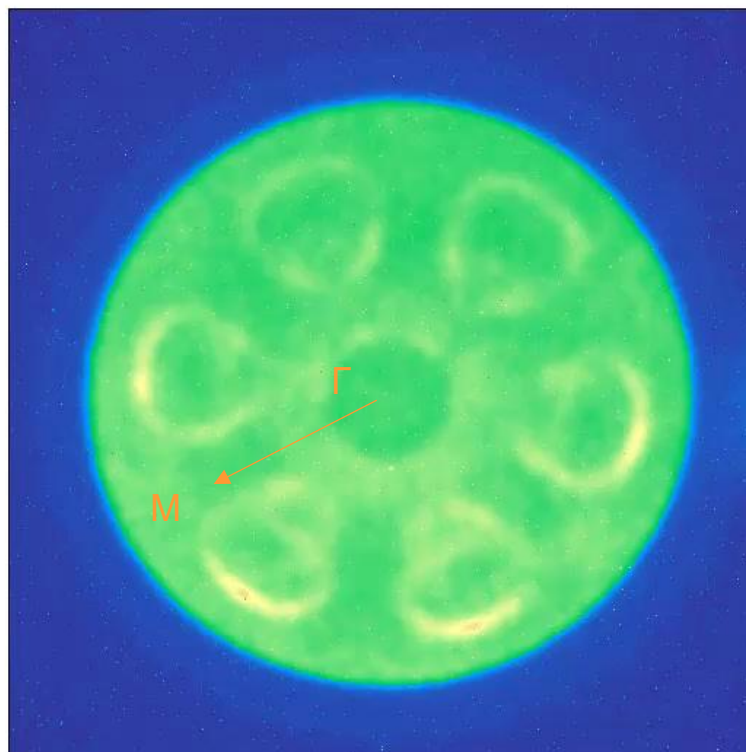
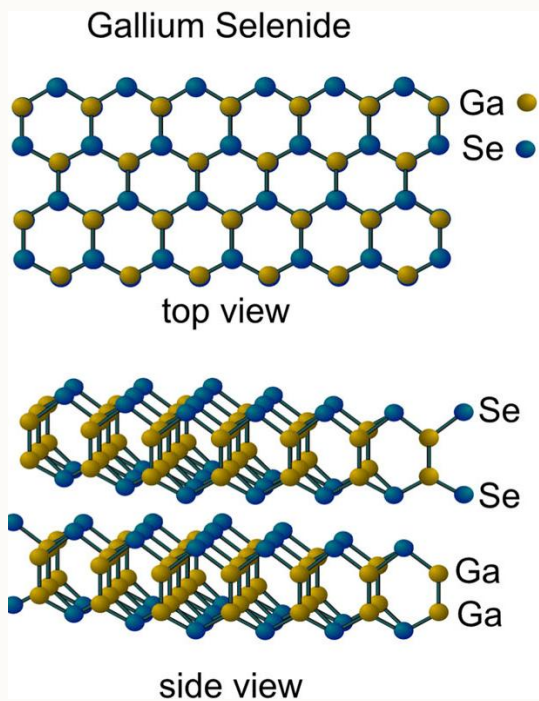


X-ray excited secondary electron image





GaSe Flake (NanoESCA/ARPES)





Epitaxy of GaSe Coupled to Graphene: From In Situ Band Engineering to Photon Sensing

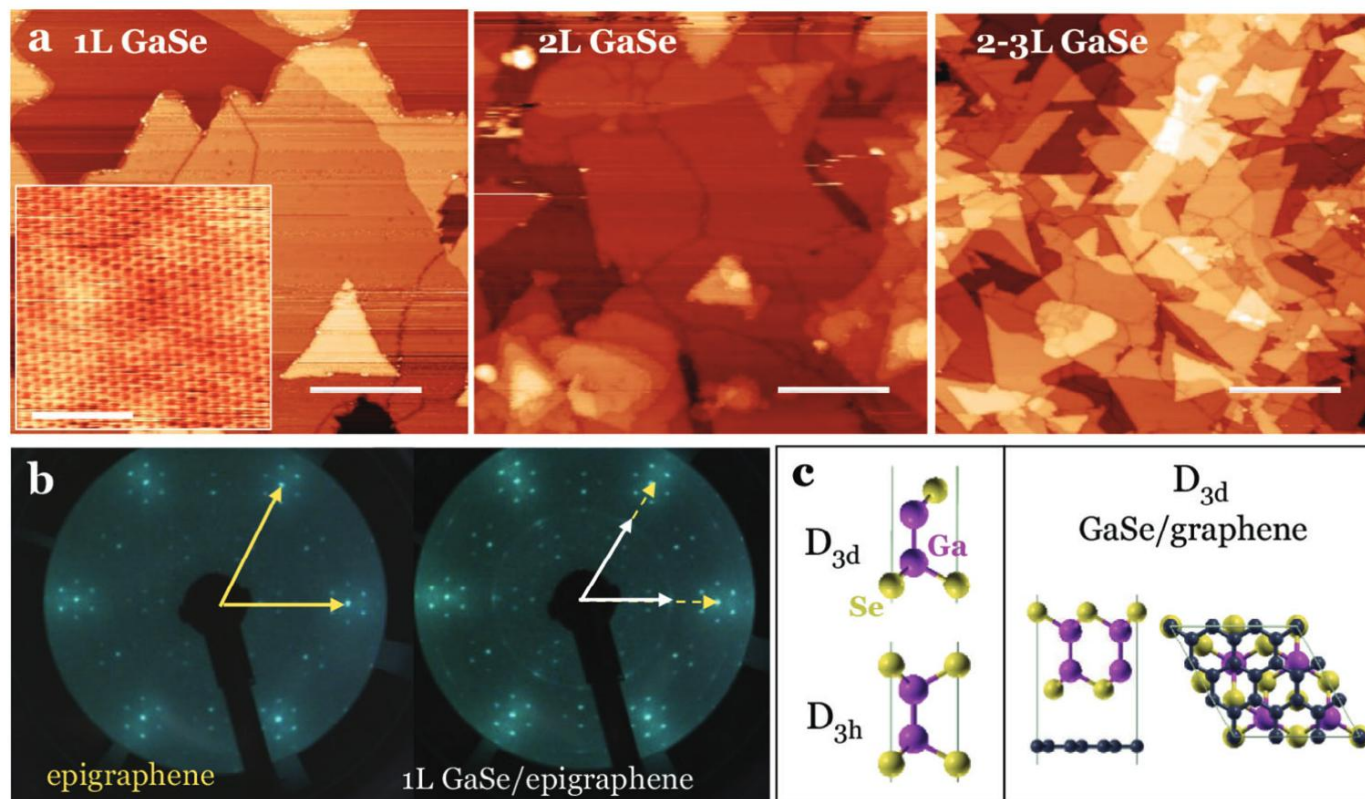
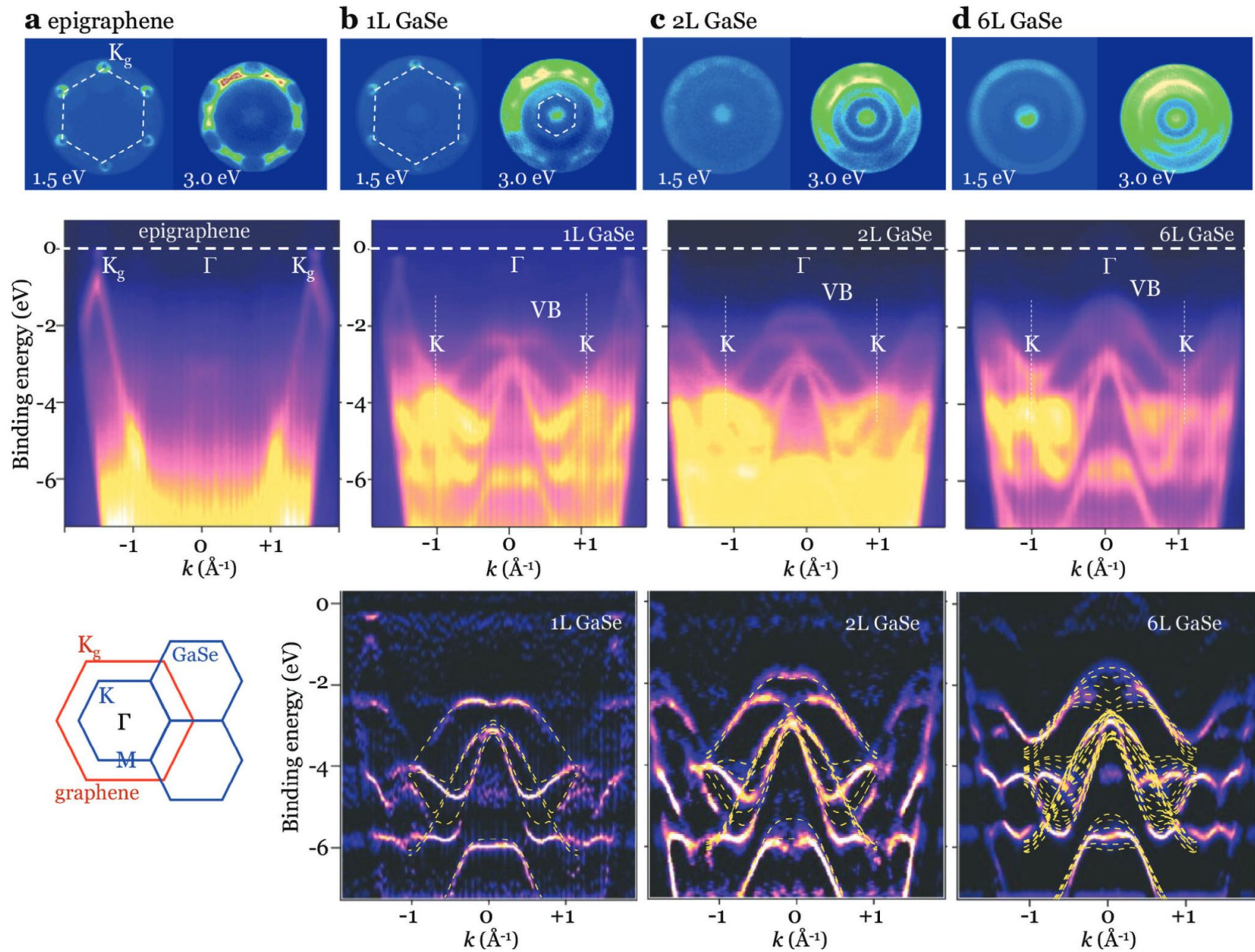
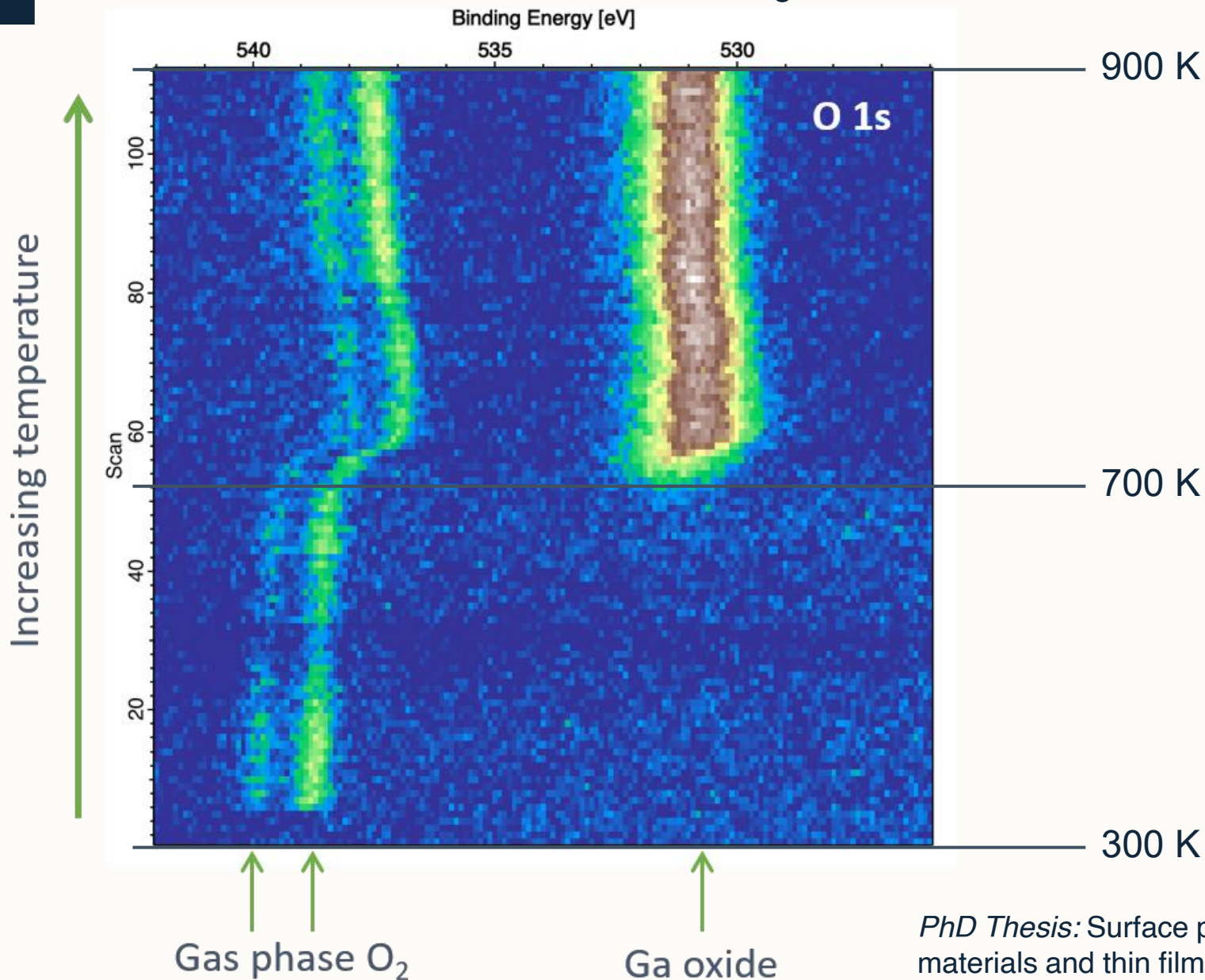


Figure 2. a) Scanning tunneling microscopy (STM) images for 1 layer (L), 2L, and 2–3L GaSe on epigraphene (scale bar = 200 nm). The inset (left) shows a high-resolution STM image for 1L GaSe (scale bar = 4 nm). b) Low energy electron diffraction (LEED) image for epigraphene (left) and for 1L GaSe on epigraphene (right). Arrows mark the diffraction spots for graphene (yellow, left and right) and 1L GaSe (white, right). Each of the spots for graphene is surrounded by six satellite spots associated with the $(6\sqrt{3} \times 6\sqrt{3})R30^\circ$ buffer layer of SiC. c) Left: polymorphs of GaSe with axial-symmetry (D_{3h}) and centro-symmetry (D_{3d}). Right: In-plane and side views of D_{3d} GaSe on graphene, as modeled by density functional theory.





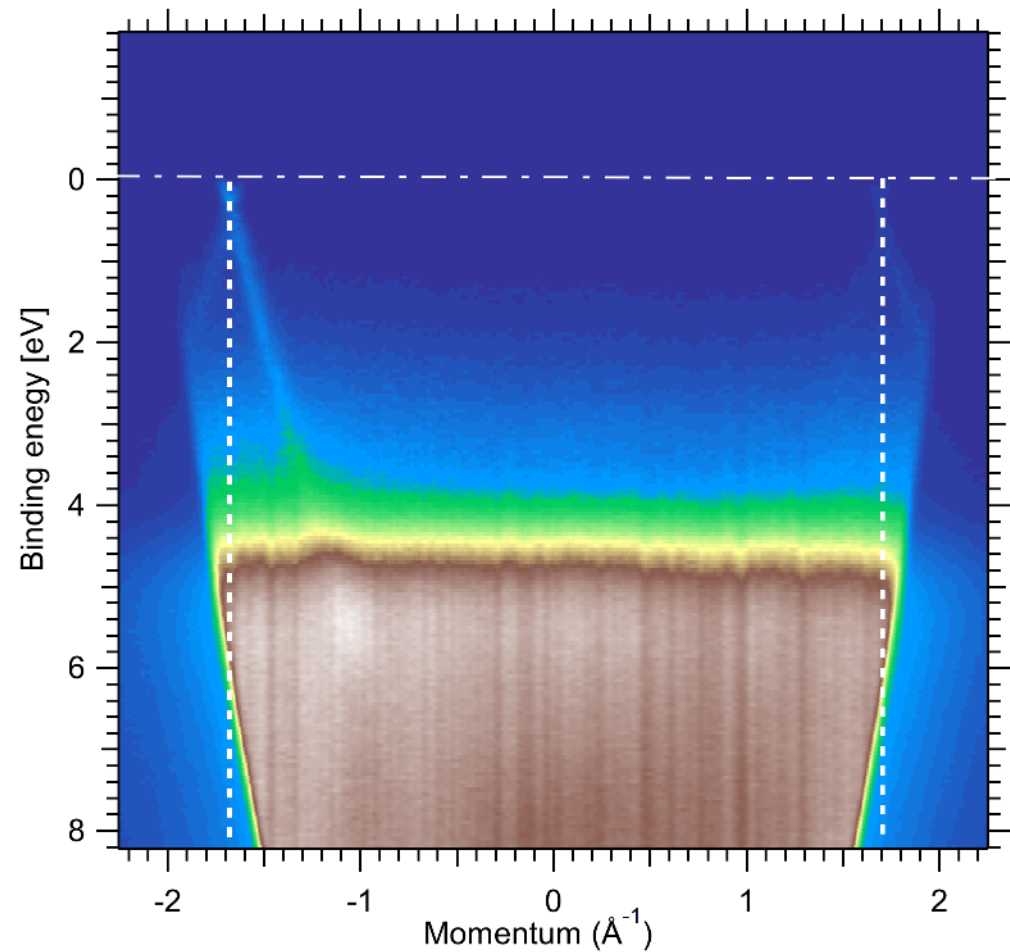
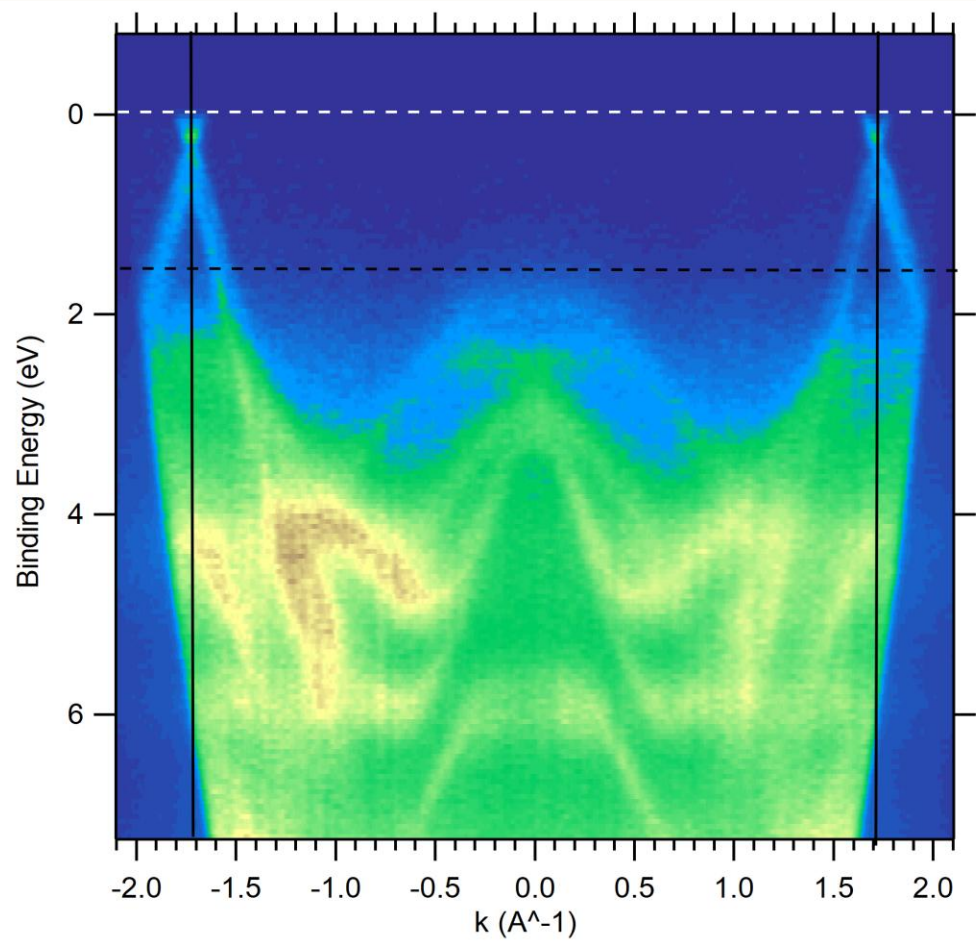
Oxidation of 3-layer GaSe to Ga₂O₃



PhD Thesis: Surface properties & electronic band structure of 2D materials and thin films, K. Rahman (2024) – University of Nottingham

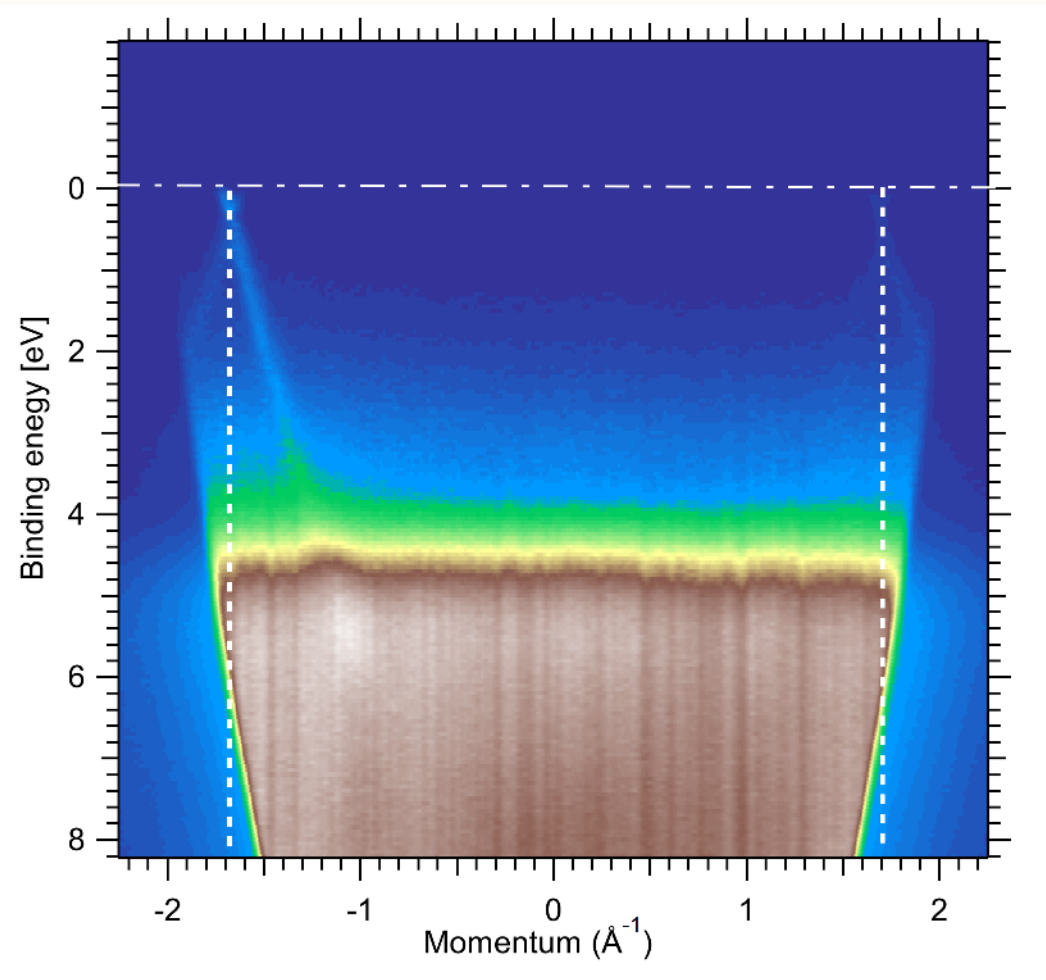
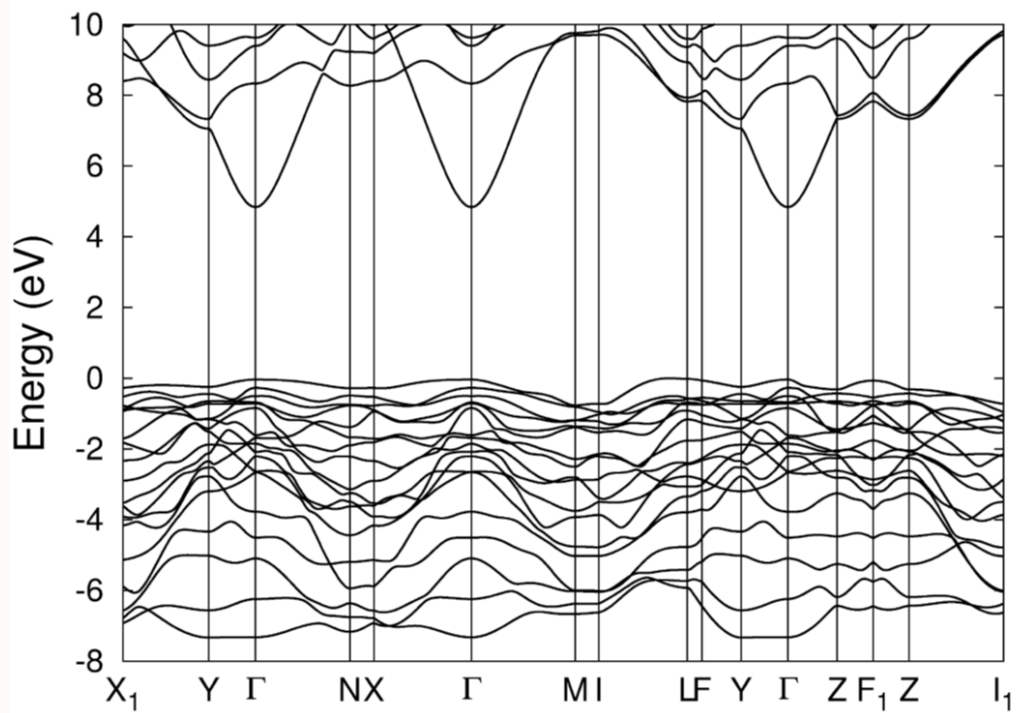


Oxidation of 3-layer GaSe to Ga_2O_3



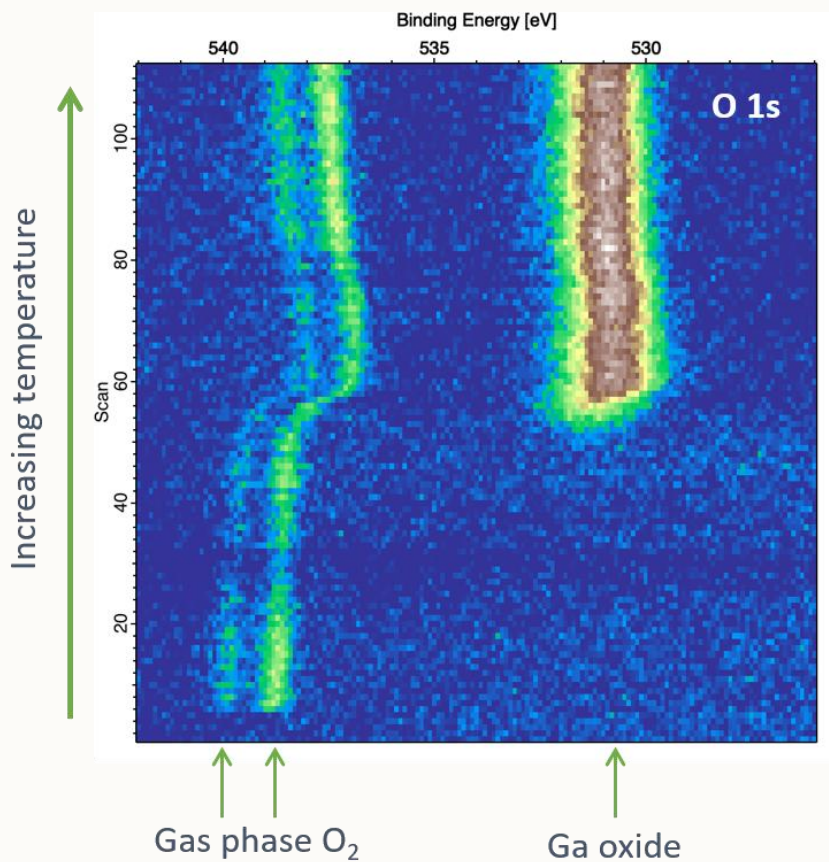


Oxidation of 3-layer GaSe to Ga_2O_3

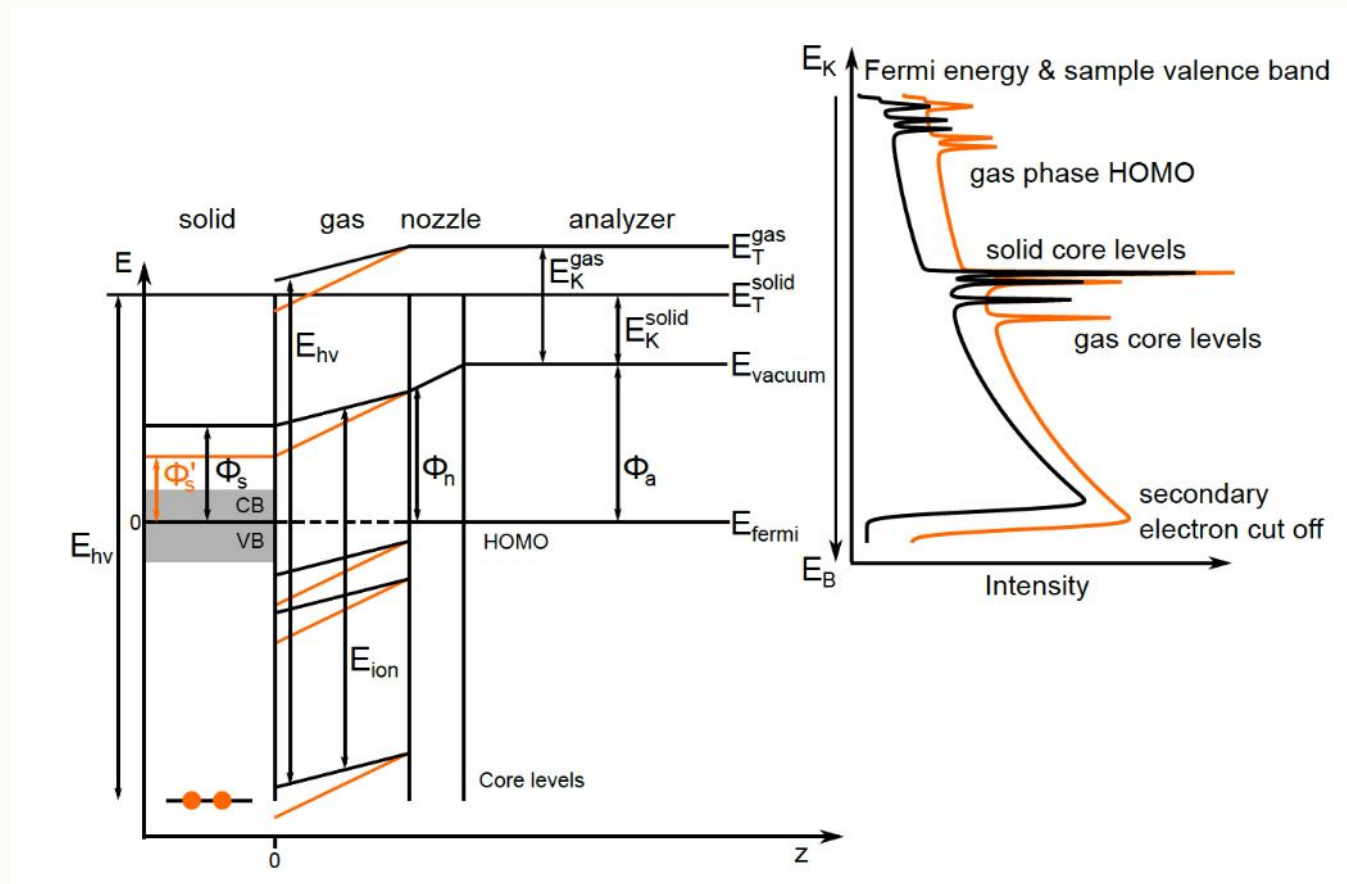




In-situ NAP-XPS probe of the work function



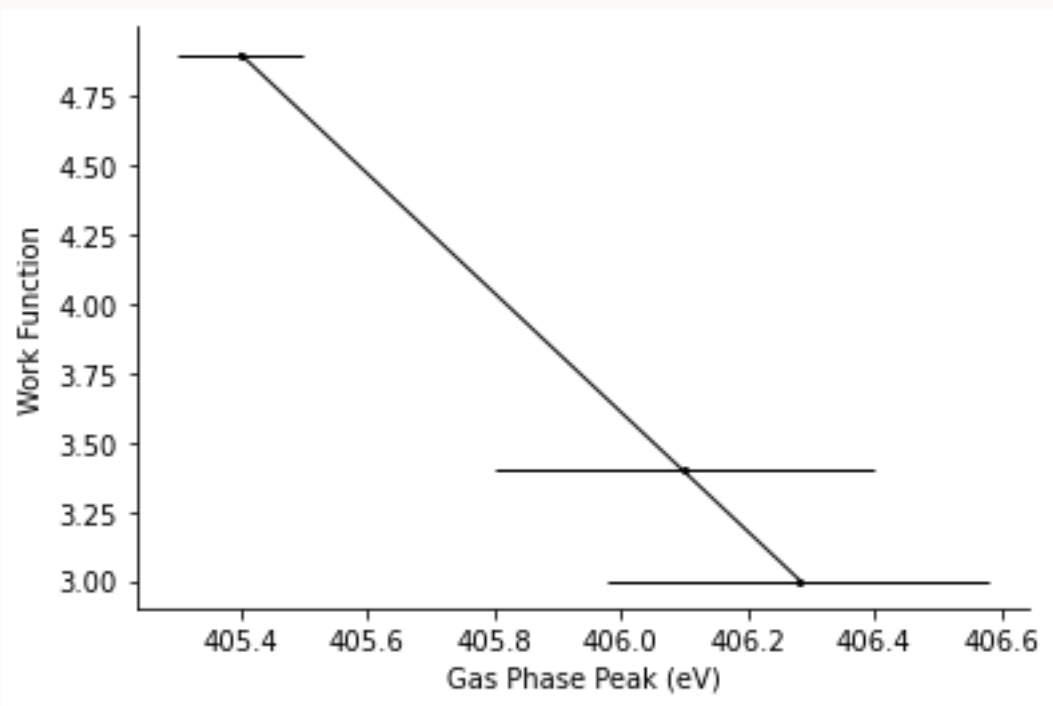
K. Rahman (2024) PhD Thesis



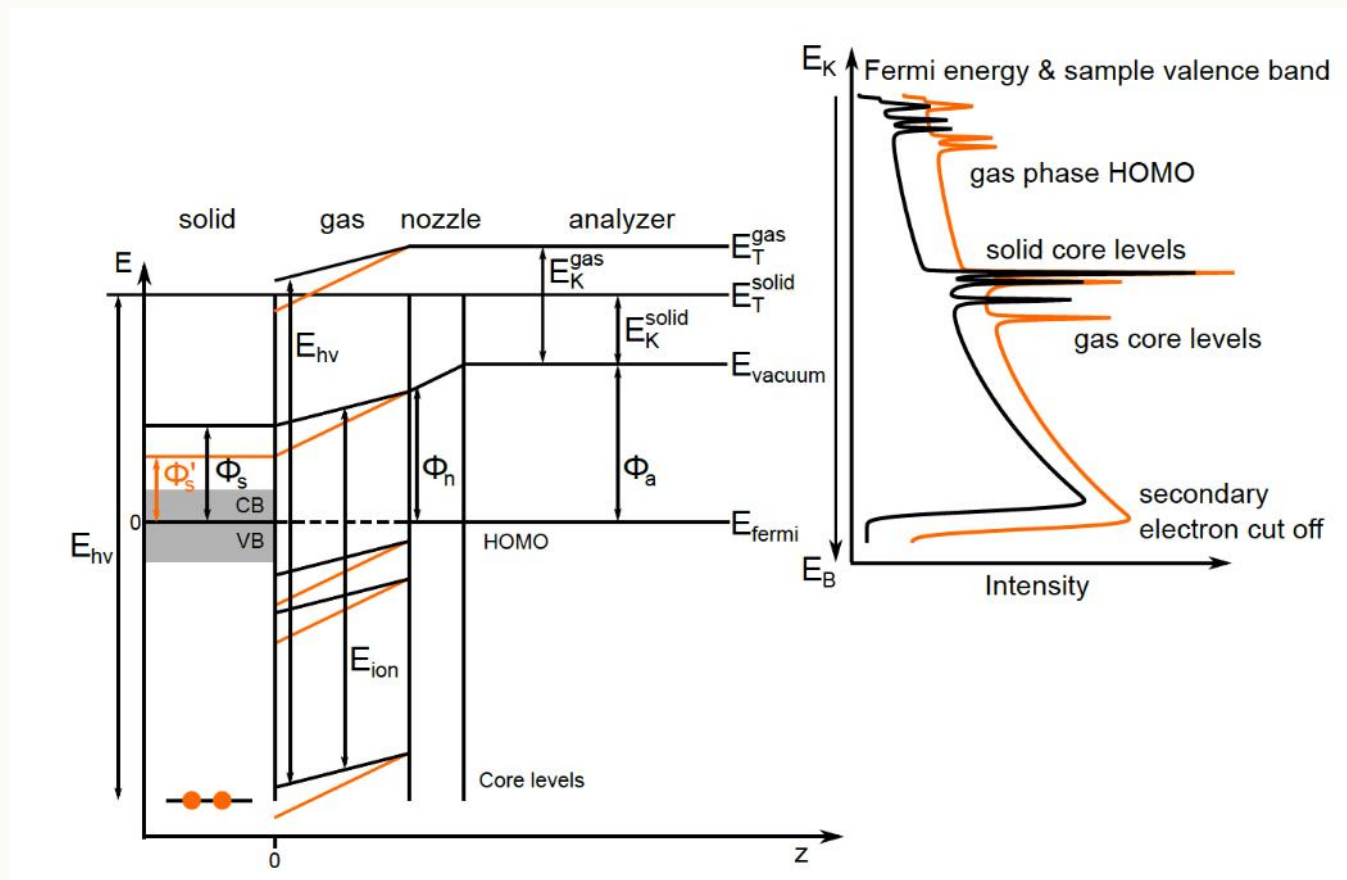
PhD Thesis: Synchrotron-based In Situ Electron Spectroscopy Applied to Oxide Formation and Catalysis J. Niclas (2017) – Lund University



Gas-phase XPS probe of the work function



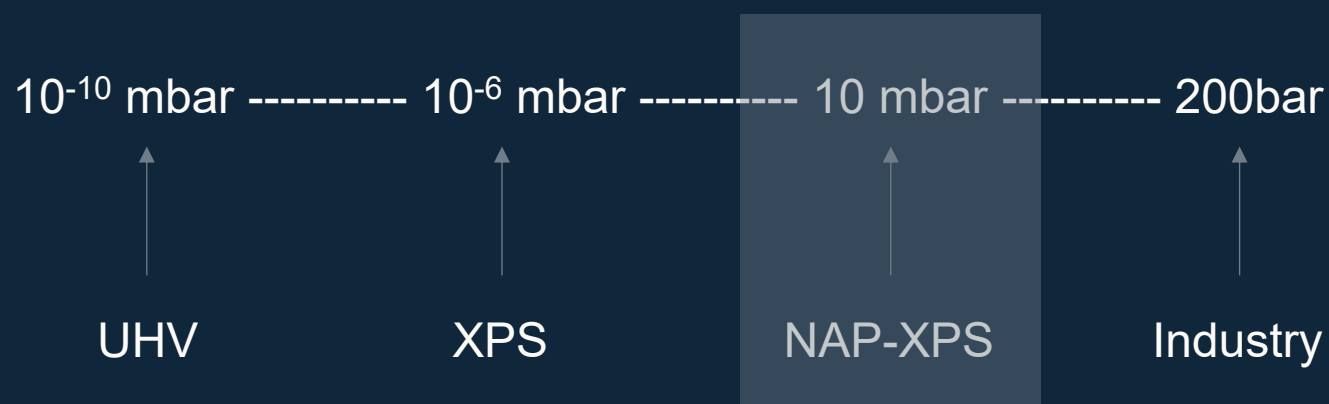
4 mbar N_2 over Au, O/Al, and Al
M. Parker, B. Linton, J. N. O'Shea (Unpublished)



Synchrotron-based In Situ Electron Spectroscopy Applied to Oxide Formation and Catalysis J. Niclas (2017) PhD Thesis



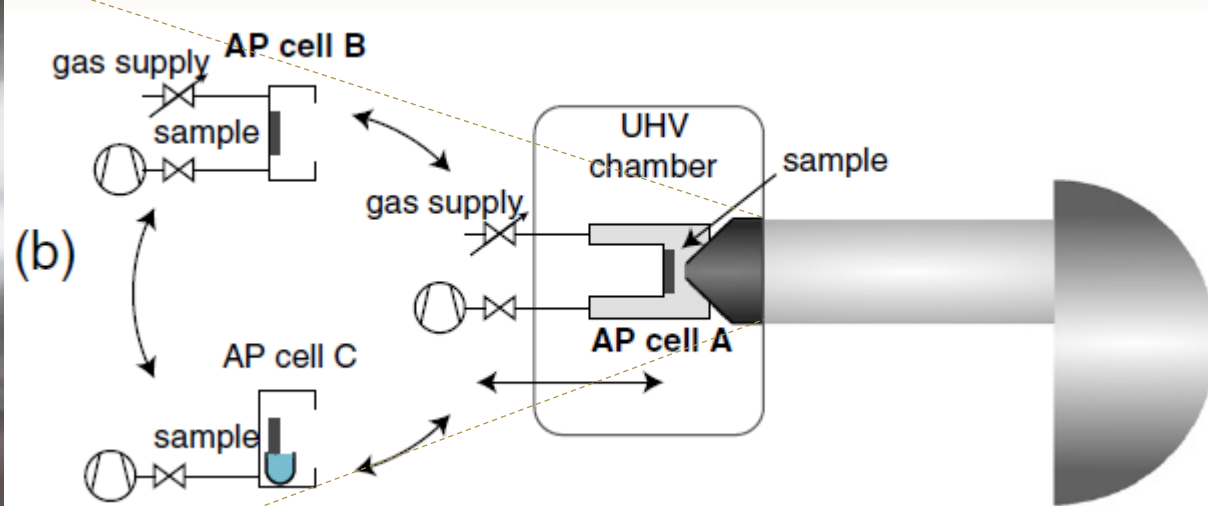
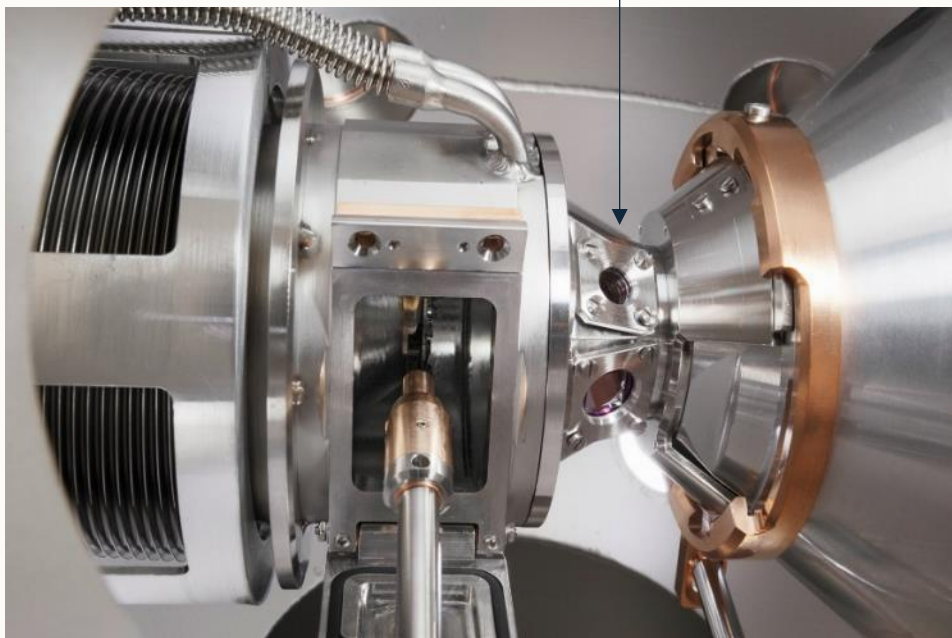
What does near-ambient pressure mean in the context of electrochemical XPS?





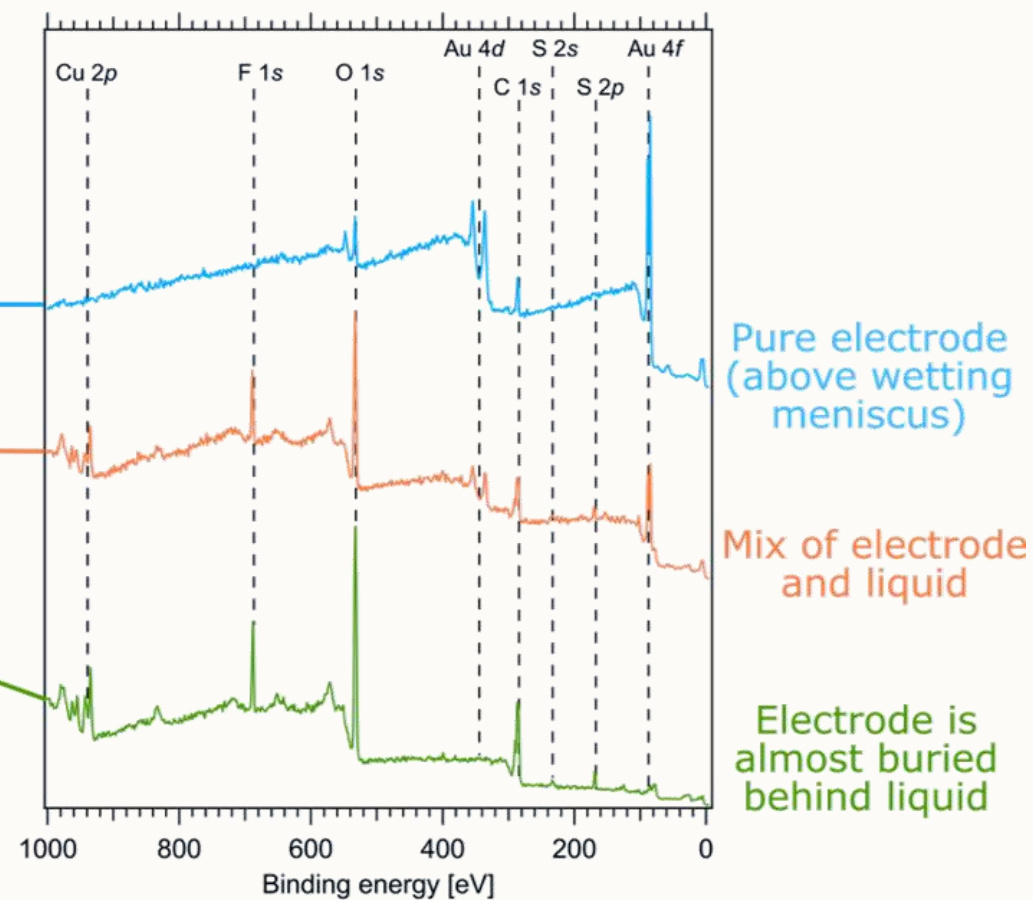
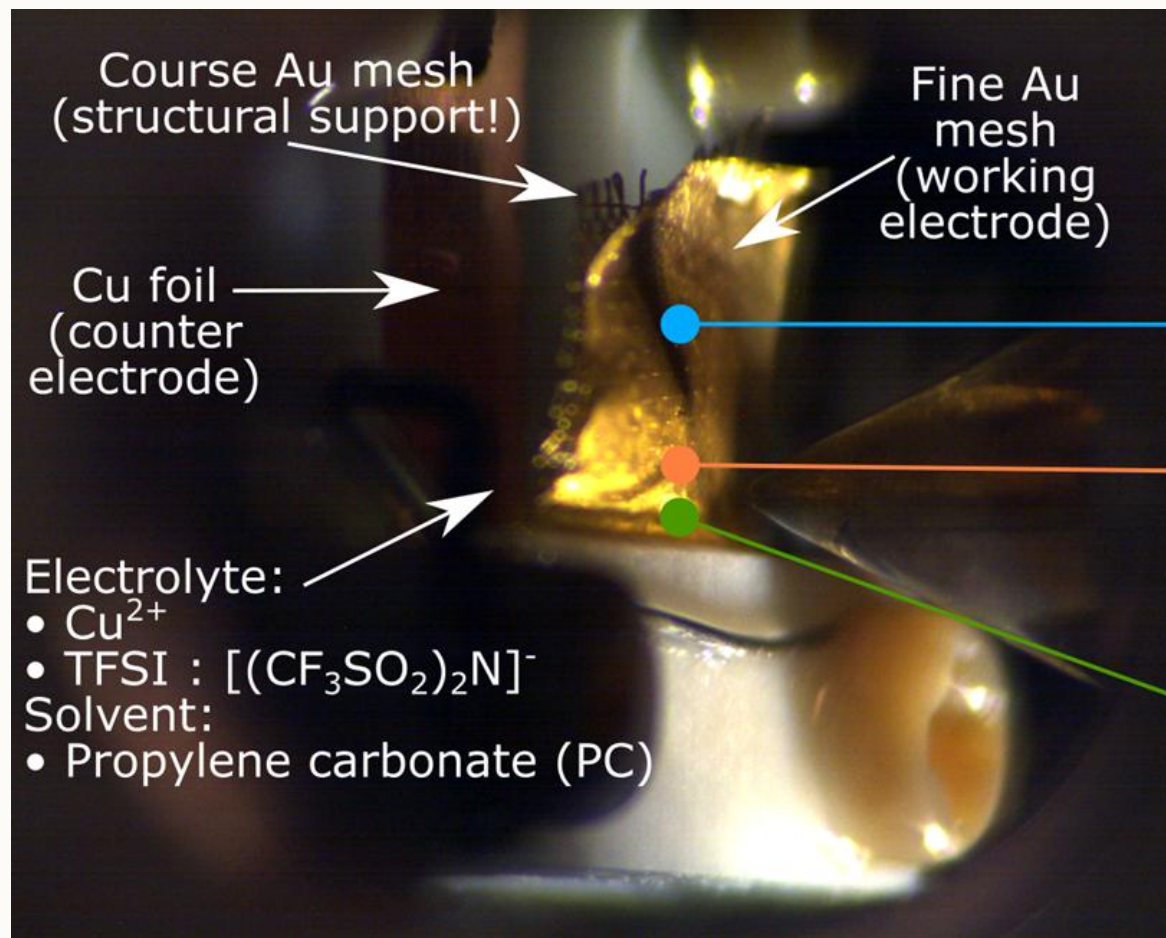
In-situ operando electrochemistry

X-rays





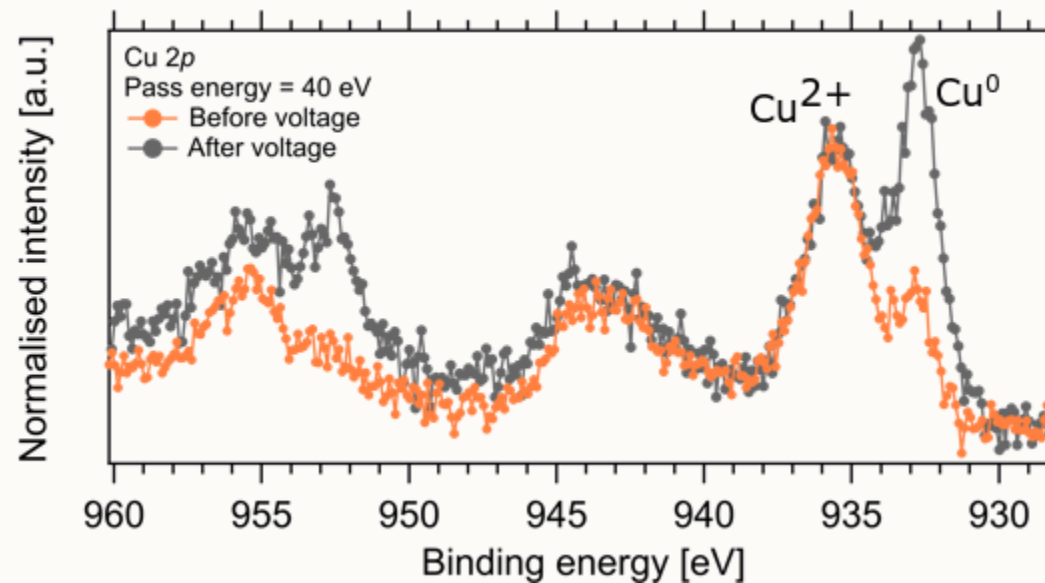
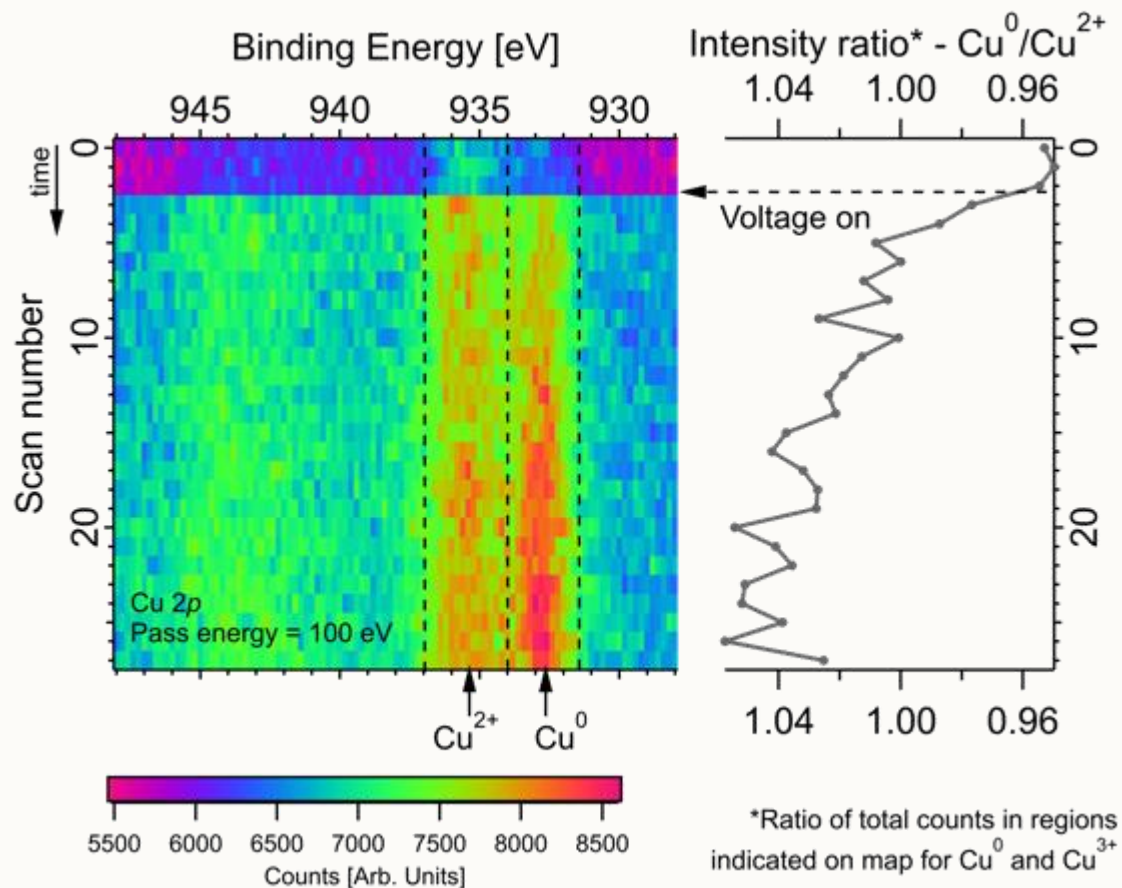
In-situ operando electrochemistry



R. H. Temperton, Unpublished



In-situ operando electrochemistry

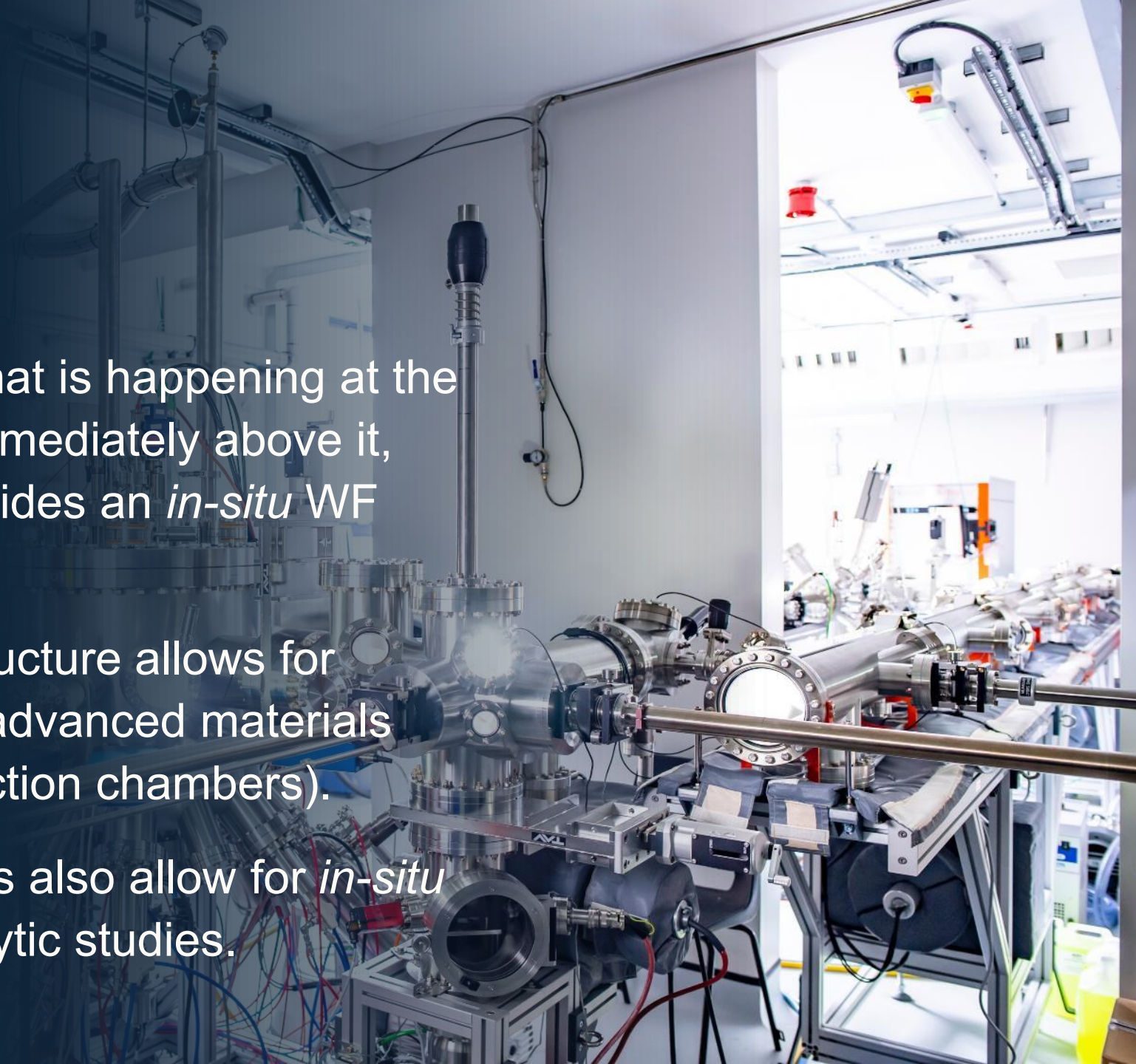


R. H. Temperton, Unpublished



Summary

- NAP-XPS allows us to see what is happening at the surface and the gas-phase immediately above it, **during the reaction** and provides an *in-situ* WF probe.
- The surrounding UHV infrastructure allows for transfer from instruments for advanced materials preparation (and coupled reaction chambers).
- Different sample environments also allow for *in-situ* electrochemical or photocatalytic studies.





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